

A STATISTICAL ANALYSIS OF HOT-CORROSION TESTS OF SOME EXPERIMENTAL AND COMMERCIAL SUPERALLOYS

R. Field, et al

Naval Ship Research and Development Center Washington, D.C.

January 1969



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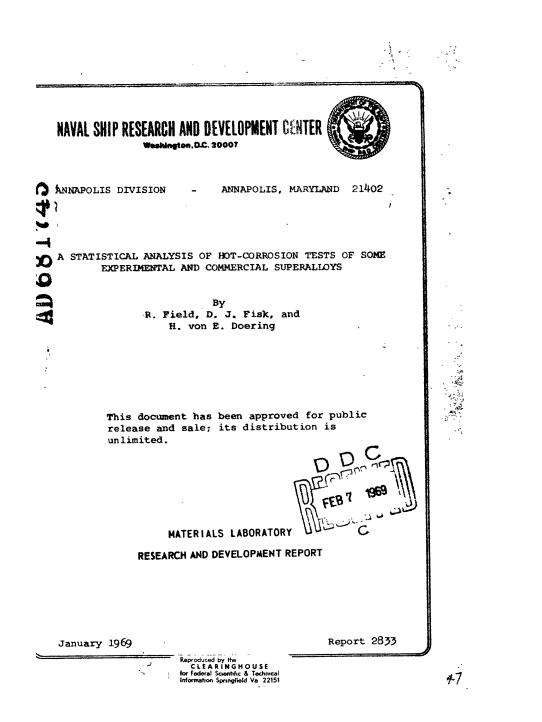
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A Statistical Analysis of Hot-Corrosion Tests of Some Experimental and Commercial Superalloys

> by R. Field, D. J. Fisk, and H. von E. Doering

#### ABSTRACT

The use of gas turbines in marine power plants depends in part on the development of superalloys which not only possess high temperature mechanical properties but also resist the corrosive effects of sea salt.

As part of a program to develop such allovs, a total of 137 experimental and commercial superalloys, both nickel and cobalt based, were exposed in burner rigs where controlled amounts of sea salt were added to the combustion products of sulfur-containing diesel fuel. Test temperatures ranged from 1600° to 2125° F. Times ranged from 86 to 100 hours with 200 parts per million, and from 489 to 1100 hours with 5 parts per million salt. Corrosion was measured by recording both surface loss and maximum penetration. This experimental work was performed by the General Electric Company under contract to the Naval Ship Research and Development Center.

For each group of alloys tested under similar conditions, a linear regression equation was found that shows the average contribution of each alloying element to the amount of corrosion. The effects of the alloying elements were found to vary with changes in temperature, salt concentration, and whether or not the particular element was part of a simple binary or tertiary alloy, or a complex alloy.

Analyses of variance methods were applied to two sets of factorially designed compositions, one of nickel-base alloys and one of cobalt-base alloys, to determine the possible significance on corrosion of various proportions of single elements and interactions among elements. It was found that in the cobalt alloys significant interactions existed between heat treatment and temperature as well as between heat treatment and chromium content.

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#### RATIVE INFORMATION

report constitutes Fiscal Year 1969 Milestone 3, on page the October 1968 Program Summary of the Annapolis Division, Ship Research and Development Center. This work was ed by MATLAB Assignment 1-815-122-A, Sub-project 06 14, Task 3888, on Gas Turbine Materials, Corrosion.

results of this study were presented at the Fall Meeting Metallurgical Society of the AIME on October 16, 1968.

#### EDGMENT

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authors wish to express their thanks to Dr. Alan U. Seybolt General Electric Research Laboratory who had originally ted that this study be made; also to Mr. Fred J. Gallagher Computer Branch for planning and executing the computer ns.

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#### NAVAL SHIP RESEARCH AND DEVELOPMENT CENTER

A STATISTICAL ANALYSIS OF HOT-CORROSION TESTS OF SOME EXPERIMENTAL AND COMMERCIAL SUPERALLOYS

> by R. Field, D. J. Fisk, and H. von E. Doering

#### INTRODUCTION

The effects of alloying elements on the hot-corrosion resistance of nickel- and cobalt-base superalloys has been a subject of interest to gas turbine manufacturers for a number of years. The application of gas turbines in marine environments has necessitated the development of alloys, for hot section components, which are resistant to the molten salts ingested by the engine. A knowledge of the behavior of alloying elements in either increasing or decreasing corrosion resistance is necessary for future alloy development.

In two recent studies of hot-corrosion resistance of superalloys,<sup>1,2</sup> statistical analysis was employed to establish a multiple linear regression equation relating the weight percent of alloying elements present with the amount of corrosion observed.

It is the purpose of this study to treat statistically the data which was generated for this laboratory in four studies, under contract with the General Electric Company.<sup>3,4,5,6,7,8</sup>

Linear regression coefficients and their significance are computed for all alloying elements used in simple experimental (up to four elements) alloys, experimental complex alloys, and commercial alloys. The effects of temperature, the concentration of sea salt, and the type of alloy on the behavior of each element have, where possible, been examined.

Since it was felt that alloying elements do not behave independently but interact, two factorially designed sets of experimental alloys are examined using analysis of variance methods.<sup>9,10</sup>

<sup>1</sup>Superscripts refer to similarly numbered entries in Appendix B.

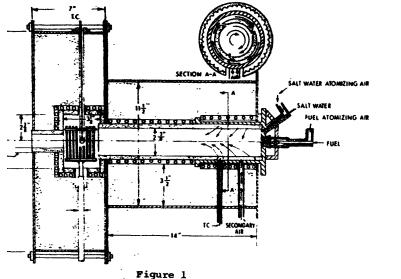
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#### EXPERIMENTAL PROCEDURE

ecimens of all 137 alloys were exposed in a burner rig d to simulate the environment within the hot section of a bine which was operated while ingesting aerosol sea salt. l is a schematic view of the equipment. Diesel fuel ing 1% sulfur was atomized and burned within a ceramic ion tube. To the flame sea salt was added at either 5 ppm ppm by weight of air.\* Tests with 5 ppm salt were run for 1000 hours, whereas tests with 200 ppm salt were run for p to 100 hours only. Thermal cycling was effected by g the rotating specimen holder and allowing the specimens for 5 minutes every 50 hours during the 500- and 1000-hour The shorter tests were not thermally cycled.

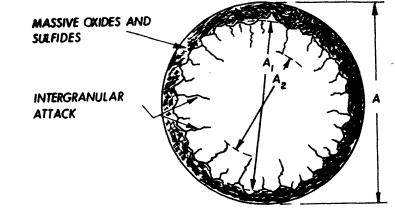
e specimens, nominally 1/8 inch in diameter and 1 5/8 inches th, were sectioned after exposure; two measurements, surface 1 maximum penetration, were taken as shown in Figure 2.



Schematic Cross-Section of Burner Rig

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lations used in this text are from the GPO Style Manual, unless otherwise specified.



- A = ORIGINAL DIAMETER, MEASURED WITH A MICROMETER.
- A1 = DIAMETER OF STRUCTURALLY USEFUL METAL. MEASURED AT 100X
- A<sub>2</sub> = DIAMETER OF METAL UNAFFECTED BY OXIDES AND SULFIDES, MEASURED AT 100X

SURFACE LOSS: A-A1 LOSS IN DIAMETER DUE TO MASSIVE OXIDES AND SULFIDES.

MAXIMUM ATTACK: A-A2 LOSS IN DIAMETER DUE TO ALL FORMS OF OXIDATION AND SULFIDATION.

Figure 2

Method of Measuring Hot-Corrosion Attack

#### RESULTS AND DISCUSSION

#### CORROSION MEASUREMENTS

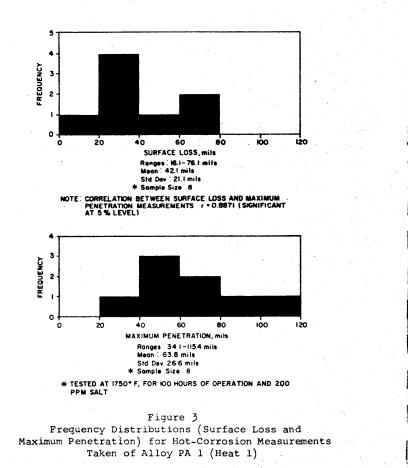
To assess the consistency and accuracy with which corrosion behavior can be measured in the burner rig used in the study, measurements were taken from eight specimens of one heat from an alloy, PA 1 (Heat 1). Each specimen was originally intended as a control to determine the similarity of nominal operating conditions between runs. Therefore, each specimen comes from a different test run, although each test was performed under the same specified conditions of temperature (1750° F), salt concentration (200 ppm) and operating time (100 hours). Thus, the

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Ince,  $\sigma^2$ , for the specimens represents dispersion in test tions as well as variation in behavior of the alloy from men to specimen.

A comparison of the data for the surface loss and maximum ration measurements indicate an average difference of about .ls. There is, at the 5% significance level, a significant ir correlation between the surface loss and maximum penetrameasurements, see Figure 3.



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Alloy PA 1 (Heat 1) was the only alloy in the study for which the results of a sufficient number of tests performed under similar conditions were available to evaluate the distribution parameters of surface loss and maximum penetration. These estimated measures of dispersion and correlation between surface loss and maximum penetration should not be applied to other alloys, however, or to PA 1 (Heat 1) when they are tested under other conditions.

#### REGRESSION ANALYSIS

An intensive regression analysis of the 137 alloys included in the development program was completed.

The alloys analyzed included 47 experimental cobalt-base alloys, 73 experimental and 17 commercial nickel-base alloys. The test temperatures ranged from 1600° to 2125° F. Times ranged from 86 to 100 hours with 200 ppm salt and from 439 to 1100 hours with 5 ppm salt. A total of 969 tests were examined in the regression analyses. The compositions by weight percent of simple experimental, complex experimental, and commercial nicklebase alloys, plus experimental cobalt-base alloys, are given in Appendix A, Tables 1-A through 4-A.\* The analyzed composition is given when available, otherwise the nominal is shown. The General Electric Research Laboratory series designated as RL nickel-base alloys are simple experimental alloys, while the Thomson Engineering Laboratory (TEL) series and the Marine Engineering Laboratory nickel (MELNI) alloys are the complex nickel-base alloys. The RL series of the cobalt-base alloys, the Marine Engineering Laboratory cobalt-base (MELCO) series, and the experimental DISCO series comprise the experimental cobalt alloys. The DISCO alloys were intended to be a matrix for dispersion strengthening.

Table 5-A shows the ranges of concentration in weight percent for each element in each group of alloys. The regression equation for the group will be valid only for an element whose concentration lies within the specified interval.

For each group of alloys within a series and tested under similar conditions, a multiple linear regression equation showing the average contribution of each alloying element to the amount of corrosion was found. Tables 6-A and 7-A give the regression coefficients for the equations representing the different groups of alloys at various conditions. Coefficients in Table 6-A are

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All tables mentioned in this text will appear in Appendix A.

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on the measurement of surface loss, whereas those in Table e based on maximum penetration measurements. A positive sion coefficient indicated a tendency for a given element rease corrosion whereas a negative coefficient indicated a sed tendency toward corrosion. Other pertinent facts ning each regression equation are the multiple correlation cient, R, and the standard error of estimate, S<sub>E</sub>.

nspection of Tables 6-A and 7-A indicates the contribution rosion of relatively few coefficients with a high level of lence (95%). The confidence level of many of the coefficients be accurately measured due principally to the few tests :ted of any given alloy. In addition, the coefficients of :lements do not consistently indicate that the element has a beneficial or detrimental effect. Also, it was not >le, with the number of tests used in the studies, 3.4+5.6.7,8mine all the possible interactions. Tables 3-A through 17-A we the behavior of each element as it may be affected by test :ature, by time and salt concentration, by whether or not lement is a constituent of a simple or complex alloy, and by >ncentration range of the element present.

The effect of temperature in 100-hour tests of the simple .-base experimental alloys (RL 1 through 9, and 20 through i shown in Table 3-A. It can be seen that Y, Zr, and Ce insistently detrimental with respect to maximum penetration i the three temperatures, and that C, Si, Cb, Mo, and W are isingly detrimental with increasing temperature. On the hand, Cr, Fe, and Co are consistently beneficial while Al ieficial with increasing temperatures; Ti is detrimental at F. The highest degree of confidence (95%) in the above isions is indicated for Ti (1675° F), for Zr (1750° F), for i Ce (1675° and 1750° F), for Cb, Mo, W, and Ta (1900° F), or Cr (all temperatures).

The effects of temperature on the behavior of elements with et to maximum penetration of the complex experimental nickelalloys (MELNI 1, 2, 4, 5, and 7 through 24) in 1000-hour are summarized in Table 9-A. It can be seen that Cr, Fe, c, Mo, W, and Re have consistent, but not highly significant, mental effects and that only Al, Ti, Y, La, and Ta seem to a beneficial effect on corrosion. Boron seems to promote sion with increasing temperature, in contrast to the effects and Cb. However, only the effect of Al and Y on corrosion i be considered highly significant (at the 90% confidence ).

In comparison of Tables 8-A and 9-A, it appears that the t on maximum penetration of Zr was consistently duplicated

6

in both the RL alloys for 100 hours and the MELNI alloys for 1000 hours.

' Thus, the projection of the behavior of elements in simple alloys under one set of conditions and ranges of concentration cannot predict their behavior in complex alloys at other sets of conditions. In addition, from Table 5-A it can be seen that C, Cr, Co, Zr, and W were present in different concentration ranges in each series of alloys. This fact may have added to the discrepancy.

A more direct comparison between simple and complex alloys is summarized in Table 10-A. Under precisely the same conditions,  $1750^{\circ}$  F and 200 ppm salt for 100 hours, consistant agreement is found in Al, Ti, Fe, Y, and Zr with respect to maximum penetration. The discrepancy in C, Cr, Co, and W can again be suspected to be due to the differing concentration ranges in both sets of alloys. The beneficial behavior of Mo at  $1750^{\circ}$  F in the MELNI series in the 100-hour test, in contrast to the 1000-hour test, or in the simple alloys with increasing temperature may indicate that this element contributes to corrosion resistance in a highly sulfidizing environment, but not in a more oxidizing one associated with higher temperatures and lower salt concentrations.

Similar comparisons can be made from Tables 8-A through 10-A with respect to surface loss. In the RL series, the beneficial (or detrimental) influence of C, Ti, Cr, Fe, Co, Cb, and Mo on corrosion was the same for both surface loss and maximum penetration. On the other hand, the influence of B, Y, Zr, and Ce on surface loss was opposite to that for maximum penetration.

A similar lack of agreement exists for the MELNI series at  $1600^{\circ}$  and  $1800^{\circ}$  F (Table 9-A). Only C, Fe, Zr, Mo, W, and Re had the same influence on both surface loss and maximum penetration. The disagreement is not surprising in view of the lack of correlation between the two measurements, and thus gives added support for the use of both measurements in hot-corrosion tests.

The test effects of time and salt concentration at  $1750^{\circ}$  F on the MELNI 1 through 13 series is shown in Table 11-A. It is interesting to note that the effect of time does not change the sign of any regression coefficient although the values do differ. Also, the influence of each element is the same for both surface loss and maximum penetration. A comparison of the effects of the simple RL alloys and the complex TEL and MELNI alloys is given in Table 12-A.

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t is interesting to note that none of the effects are cont, either for surface loss or for maximum penetration. The s on corrosion for the elements common to the three groups s the same for both types of measurements.

similar comparison is made between the TEL alloys and the cial alloys tested at 1600° F, for 500 hours with 5 ppm Table 13-A). Of the elements common to both series of , all but C, Ti, and Cr show the same beneficial, or ental, effect. With the exception of W, all elements had me effect with respect to both surface loss and maximum . With regard to the surface loss of the commercial alloys, fects of Cr, Mo, and W agree with results of Ryan<sup>2</sup> for tests med at 1800° F.

n contrast to the simple nickel-base RL series, the cobalt (RL 10 through 15 and 52 through 64) in Table 14-A show loying elements to have the same type of effect on both e loss and maximum penetration. Carbon is always ental, while Ni, Mo, and W show an increase in corrosion with increasing temperatures. Titanium, Cr, Y, Zr, Cb, La, d Ta show a beneficial effect with increasing temperature.

The behavior of alloying elements in the complex MELCO 1 in 11 series is compared with their behavior in the simple ties at two temperatures,  $1750^{\circ}$  and  $1900^{\circ}$  F, for the 100- and 200 ppm salt tests. The elements common to both sets of i exhibit a lack of consistent behavior, with the exception whereby the set of alloys used and the temperature tested have an effect on surface loss and maximum penetration. Im shows a beneficial effect (sometimes significant, somenot) under all the conditions described above.

The effects of temperature on the behavior of elements common ICO 3, 4, 5, 7, 3, 9, 10, and 11 can be examined in Table Boron is consistently detrimental at all temperatures and th surface loss and maximum penetration, although the sigance is questionable. Chromium, Ni, Ta, and W show reduction crosion with increasing temperature, and C, Cu, and Y show sistent test results. However, because of the few tests ted on each alloy, the validity of these conclusions is ionable.

The nature of the tests conducted on MELCO 1 through 11 ts us to examine the behavior of the alloying elements in nat more detail than can be done with most of the other s examined. In addition to the sign of the regression icient, its numerical value may be examined in Tables 17a-A

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through 17h-A to determine the sensitivity to corrosion of a given element under two test-times and two temperatures.

#### Boron

With the exception of 1000-hour tests at  $1750^{\circ}$  F, B seems to be detrimental with respect to surface loss. Maximum penetration is more dramatically enhanced at 1900° F than surface losses at that temperature, even with the lower salt concentration.

#### Carbon

The effect of carbon appears to vary most with test conditions at 1900° F, although the significance of the coefficients is questionable.

#### Chromium

Although Cr is accepted as beneficial in hot corrosion, it appears from the tests that there are cases at the high concentration ranges where it is not. For example, based on maximum penetration values, Cr appears beneficial at both 1750° and 1900° F with the higher salt concentration and 100 hours of operation. However, a longer period of test operation (1000 hours) shows a detrimental effect at a low salt concentration. This seems to indicate that test conditions, oriented more toward oxidizing than sulfidizing conditions, cause Cr to be less helpful.

#### Nickel

Nickel appears to be the most detrimental at the highest temperature and longest test even with the low salt concentration.

#### Copper

Copper appears to be beneficial, expecially at high temperatures and long times, with the exception of the test at  $1750^{\circ}$  F for 1000 hours.

#### Yttrium

The effect of Y at all conditions, except the detrimental effect on surface loss in 1000-hour tests, is beneficial. These observations, at least with respect to maximum attack, support the contention that if Y is helpful in oxidation resistance of cobalt-base alloys, then it should be of similar help in corrosion resistance.

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#### alum

In the concentrations present in MELCO 1 through 11, Ta ars to be innocuous (neither beneficial nor detrimental).

#### sten

Like Ta, W, too appears innocuous or at least inconclusive in behavior.

#### YSIS OF VARIANCE

#### cel-Base Alloys

Table 18-A illustrates the testing of three sample specimens sight nickel-base alloys at each of three temperature levels total of nine specimens per alloy).

The analysis of variance (Table 19a-A), using all the given : results for maximum penetration, seems to indicate that only ferences in the proportion of chromium used (15% or 20%) made ignificant contribution to differences in corrosion measurets.

However, a further examination of the data in Table 13-A icates that certain test runs (e.g., Nos. 3 and 8) resulted in exaggerated corrosive effect on certain alloys, as compared other specimens of the same alloys tested under the same ditions but in different test runs.

Examples in Table 18-A of these particular test specimens lude Alloy RL 46, when tested at 1675° and at 1750° F, and oys RL 47 and 51, when tested at 1750° F.

Therefore, a new analysis of variance was performed (Table -A) in which Tests 3, 8, and 13 were eliminated; the remaining t measurements for each alloy obtained under similar conditions e averaged together. In the new analysis, besides the ferences in corrosion due to differences in chromium, those cerning titanium and temperature levels also seemed to make nificant contributions to corrosive behavior.

It should be noted that variations in the proportion of kel used in these nickel-base alloys, although not specifically ted in Table 18-A, may also contribute to an increase or rease in corrosive effects.

#### Cobalt-Base Alloys

Hot-corrosion tests were performed on two specimens each of nine cobalt-base alloys, at each of three temperature levels, as shown in Table 20-A. For each two specimens, one specimen was placed in the test chamber as cast, while the other specimen was given a special heat treatment before being placed in the test chamber.

An analysis of variance using the maximum penetration measurement is presented in Table 21-A to determine the relative effect on corrosion of heat treatments or lack of heat treatments versus the proportions of chromium, nickel, and tantalum used within each alloy. The results of these analyses are presented graphically as well as in analysis-of-variance tables, according to statistical methods described by Hoel<sup>9</sup> and Brownlee.<sup>10</sup>

Taking into consideration all the given factors, there seems to be considerable interaction between (H) (heat treatment or lack of heat treatment) and the test temperatures used (T), as well as between (H) and the proportion of chromium used in an alloy (Cr). A significant interaction indicates that a specific combination of factors may possibly affect the outcome of a test in a somewhat different fashion than would each of the factors considered by itself (called a main effect). This analysis also indicates a possible interaction between test temperatures and the proportion of chromium used, as well as significant main effects of temperature and nickel considered independently.

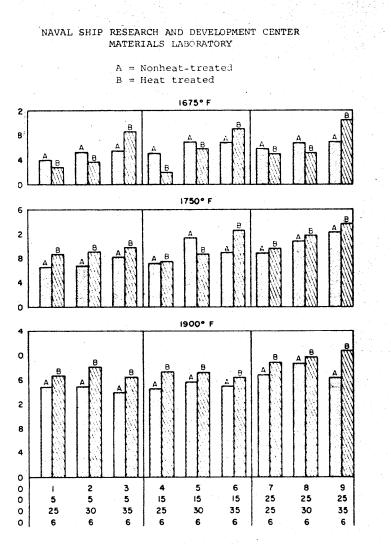
Because of the highly significant effect of temperature on most of the other factors, a separate analysis of variance was performed in reference to each separate temperature level ( $1675^{\circ}$ ,  $1750^{\circ}$ , and  $1900^{\circ}$  F). The result of this analysis is also presented in Table 21-A and shown graphically in Figures 4 and 5.

As can be readily seen on examining Figures 4 and 5, there is an increase in corrosion due to higher test temperatures for all proportions of nickel and chromium used in the alloys and in respect to both the heat-treated (B) and nonheat-treated specimens (A).

At an operating temperature of 1900° F, heat-treated specimens (B) show a consistent increase in corrosion over nonheattreated specimens (A). At 1750° F, one set of specimens out of nine sets shows a reverse relationship (a decrease in corrosion for the heat-treated specimen), while heat-treated specimens tested at 1675° F show about a 50-50 chance of either increasing or decreasing the corrosive effect.

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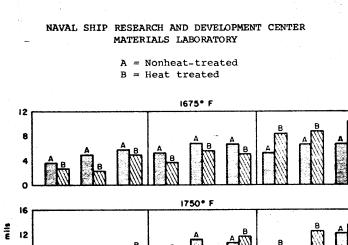
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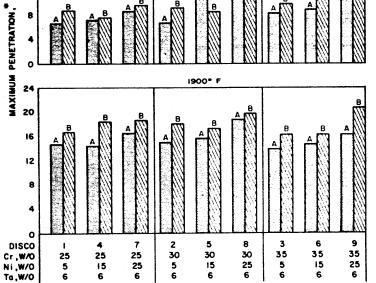


:imum penetration values from Table 20-A.

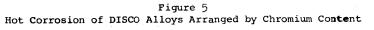
Figure 4 : Corrosion of DISCO Alloys Arranged by Nickel Content

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\*Maximum penetration values from Table 20-A.



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1 a similar fashion, the general increase in corrosion test 5 due to increases in nickel content (Figure 4) at each 1 ture level is evident, as is also the general decrease in 1 on measurements due to increases in chromium content 2 5) at each temperature level. However, an increase (or 3 in nickel content has less apparent effect on corrosion 2 sted at 1900° F, while a change in chromium content has 2 parent effect on corrosion when tested at 1675° F. The 3 of nickel, chromium, and temperature on corrosion test 3 may thus be considered main effects at various tempera-2 vels. As shown in Table 21-A, nickel seems to be very 1 cant at all three temperatures, while chromium is most 1 cant at 1750° F and least significant at 1675° F.

hus, the analysis of variance presented in Table 21-A and aphical representations in Figures 4 and 5 indicate that reatment has no significant effect on corrosion when ens are tested for 100 hours at 1675° F, a somewhat greater icance when tested at 1750° F, and a very significant effect ested at 1900° F. However, the direction of change ase or decrease in test results) versus the change in test ions is not presented in an analysis of variance, but must ermined by further examination of the data.

n addition to changes in corrosion test results due to s in chromium and nickel content, the experimenter should glect the possible effects of tantalum (6%) and the ses or decreases in cobalt content which depend on the proportion of other metals used in each alloy.

#### SUMMARY AND CONCLUSIONS

he hot-corrosion behavior of a number of experimental and cial nickel- and cobalt-base superalloys was statistically ed. Tests were made in a burner rig using diesel fuel to controlled quantities of sea salt were added. The icant results of the study are as follows:

• In general no one-to-one correspondence could be between surface loss and maximum penetration for all the tested under various conditions. Therefore, in studies such a spread of alloy compositions, both measurements are ended.

• Projection of the behavior of elements in simple under one set of conditions and ranges of concentration always predict their behavior in complex alloys at other of conditions.

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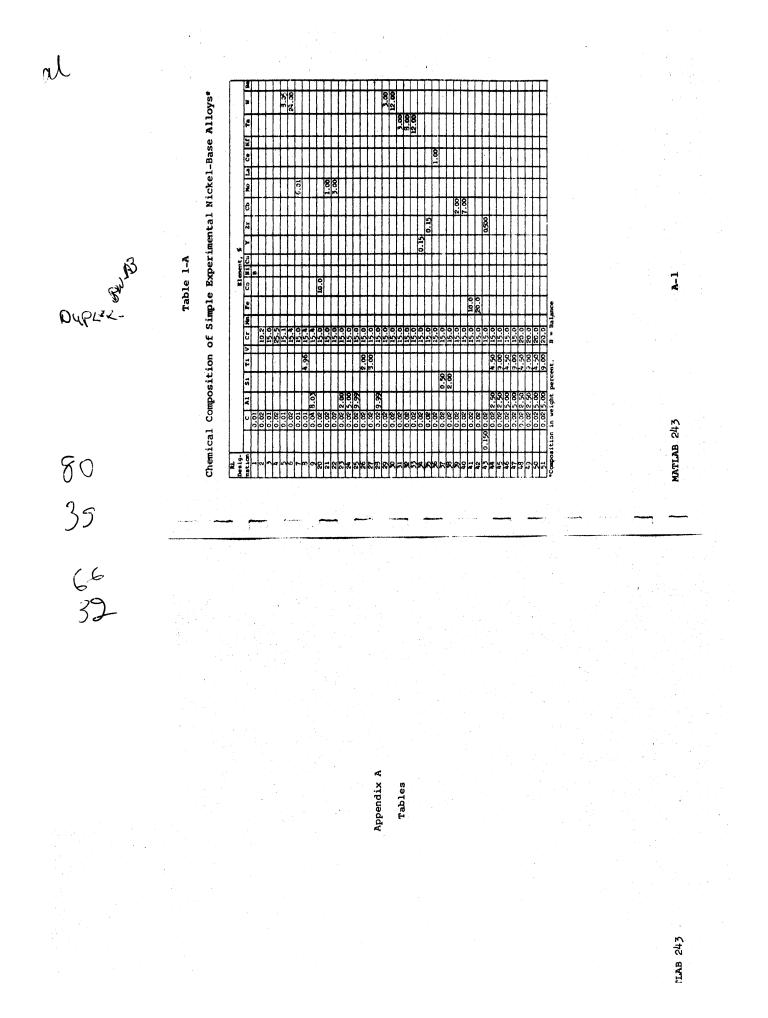
• Although Cr is considered a beneficial constituent, this analysis indicates that additions greater than about 20% in nickel-base MEINI alloys had a tendency to increase corrosion slightly.

• The lack of consistency in the effect of various elements on corrosion indicates the possible existence of interactions among elements as well as the need for testing a greater number of specimens of any given alloy.

• In a series of cobalt-base DISCO alloys with factorially designed compositions, it was found that there were significant interactions between Cr and temperature, between heat treatment and Cr, and between heat treatment and temperatures.

 Alloying elements, which show effects of doubtful significance, should be investigated by designed experiments, using a sufficient number of replications.

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#### Table 3-A

1. 1

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Chemical Composition of Commercial Nickel-Base Alloys\*

										mt,												
Designation	8	C	A1	81	TI	v	Cr	Mn	Po	Ca	MI	Cu	Y	Zr	Ср	No	La	Co	Ħť	Ta	×	Re
ANS 5397	0.014	0.18	5.50		4.70	1.00	10.0			15.0	ва			0.060		3.0						
PA 2	0.015	0.14	4.86		1.80		8.8		0.3	10.2					1.20						12.30	
ANS 5391A (Bost 1)	0.007	0.12	5.69		1.01		13.4		1.7						2.37	3.7				0.17		
aas 5399		0.08	1.60		3.15		19.1			11.1						9.9						
MS 5384 (Heat 1)	0.005	0.08	3.02		3.05		19.4									3.9						
2015 5384 + Ce	0.005	0.08	2.97	Γ	2.98		18.8		0.2	19.4	ĺ					4.2		0.50				
(Beat 2)	0.006	0.07	3.00		3.00		18.5			18.5						4.0						
PA 7 (Heat 1)	0.013	0.12	4.70		3.63		14.8			18.9	T	1		0.050		5.0						
PA 7 (Heat 2)	0.027	0.09	4.35		3.18		15.2			18.6		Γ				5.0						
PA 7 (Heat 3)	0.015	0.12	4.28		3.48		14.6			15.2	Γ	Γ				4.1					1	
PA 7 (Heat 4)	0.015	0.12	4.27	Γ	3.41		14.8			15.1	Γ					4.2	ļ					
PA 7 + 0.5 Ce	0.020	0.07	4.30		3.35		14.8			15.0	I	1		0.080		4.2	1	0.50	1	;		
PA 7 + 0.2 Ce	0.020	0.07	4.30		3.35		14.8			15.0					l I	4.2		0.20	k			
PA 1 (Heat 1)	0.014	0.10	4 .48		2.07		15.2		0.2	24.8				r t		4.6	ļ	,	[			
PA 1 (Heat 2)	0.014	0.08	4.50		2.51		15.0			25.8	1					4.5		1	2			
PA 5	0.015	0.07	5.40		2.50		11.0			15.0	Ĺ				0.45	6.5	[	l 1			1.50	
ра б	0.019	0.05	6.30		0.10		16.7		-		Γ	ļ		0.100	1.00		!	!		1.90	1.95	

\*Composition in weight percent. PA = Proprietory alloy BA = Balance

Table 2-A

1, 28 (s. 11)

1 Composition of Complex Experimental Nickel-Base Alloys\*

									1.0	ien t										
C	A1	Si	TI	V	Cr	Man	P.	Ċo	Ni	CL.	Y	Zr	СЪ	No	La	Ce	Rf	Ta	W	Re
				_					TEL											
0.12			2.00		25.0			10.0					0.50						4.00	
0.14			2.00		20.0			10.0					0.50						4.00	
0.14			1.65		15.2			10.0					0.51						4.14	
0.15			2.00		10.1			10.0					0.50						4.00	
0.16			0.02		15.1			10.0					0.50						4.00	
0.17			2.96		15.0			10.0					0.50						4.00	
0.12			3.96		15.0			10.0					0.50						4.00	
0.11			2.00		15.0			10.0					0.50						7.49	
0.12			2.00		15.0			10.0					0.50						1.77	
0.15			2.00		15.0								0.50						4.00	
0.15			1.98		15.1			24.2					0.50						4.73	
0.13	6.20		3.95		15.0			0.1					0.50	1.7						
									ZLA.	Ţ										
0.09			1.50		18.7			14.9				0.092		4.0						
0.09			1.50		18.3			14,8				0.081								4.5
0.10			2.08		20.8			15.8				0.059		0.1						
0.10			1.92		18.7			14.9	L			0.095		4.0						
0.10			2,16		18.8			23.4	Г		0.200	0.049		3.9						
0.10	3.68		1.52	Г	19.8		0.2	15.8				0.091		4.1	0.05					
0.05	2.24		3.12	Ľ	24.8			11.5	1			0.068			0.25				6.00	
80.0	1.90		2.66	Т	24.5			13.8				0.084			0.26				6.10	
0.08	2.00		4.13		22,0			13.0		<b>—</b>	1	0.063			0.12	1		2.40	5.90	
0.08	1.94		3.54		24.5			12.3		-		0.083			0.31			_	5.90	
0.05	2.20		3.10	T	25.0		-	12.3		-	1	0.084	1	-	0.17				6.00	
0.05	2.00		3.10	Т	24.8		-	12.3	1	<b>—</b>	1	2.093			0.67				6.00	
0.05	2.18		3.10		24.4		-	12.7	1	-		0.091			1.45	1			5.90	
0.14	2.26	1	4.22	Т	19.8			9.4	1	-		0.100	-		0.16	1			5.65	
0.14	2.25	1	3.93	1	19.8			3.4	1	-	1	0.015	1	2.0	0.17			1.50	1.95	
0.14			1.95		19.7		-	9.4		-	1	0.140	1		0.17				5.75	
0.12	2.50		4.40		19.5			9.6	1	1-	1	0.094	1	3.9	0.17	1				
0.14			2.54		18.5			7.9		1-	t	0.160	1		0.16				2.05	
	3.08		2.90		19.2				+	-	<u> </u>	0.120					-		3.20	
	3.00		1.37		13.3			7.6	+			0.090						1 30	1.30	

: in weight percent. BA = Balance

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#### Table 4-A

#### Chemical Composition of Experimental Cobalt-Base Alloys\*

	1 51	Ti	V C	Cr	Mn	Pe	Co	Ni	Cu	Y	Zr	Cb	No	La	Ce	Hf	Ta	W	T
								. 1	u.					•		•			
0.01			2	5.6			BA												T
0.02		į										· · · · · ·							I
0.03		Į		5.7														7.8	4
0.01		<b></b>		6.2 5.0	1						·	ļ	6.0				7.09	<u> </u>	ŧ
0.02	+	ł	1	5.4 6.4				10.2			·	ļ					7.00	<u> </u>	ł
0.45		t		5.0				10.0											t
0.45		1		5.0				10.0											ŧ
0.45	+	t	1	5.0				10.0											t
0.45	1	1	2	5.0				10.0									8.00	_	t
2.451	1		1 15	5.0				10.0									15.00		T
0.45	1.			5.0				10.0										8.00	1
0.45		<b>.</b>	12	5.0		-	_	10.0				L						15.00	4
0.45	+	\$.00	1-12	5.0				10.0		0.150		ļ						سنشبط	ŧ
0.45	+	+	1 15	5.0				10.0		0.150	····-			0.15	· · · ·	<u>  </u>		j	╋
0.45	-+	t	1 15	2.0 5.0			-	10.0				<u> </u>		<u></u>	1.00	h			t
0.45		<u>+</u>		5.0		-	-	10.0			0.500							<u> </u>	t
0.45		1.	1 12	5.0				10.0				A.00						·	t
		· · ·			_											· · · · ·			7
									LCO			1							
15 0.25	T	ĩ	1 2	7.3				10.7										7,00	3
12 0.25	1	1		0.0				0.5		0.150						1.1		0.10	
12 0.26	-	L		0.7			_	9.7									1.50		
14 0.27				9.8				9.8		1 1 20	· · · · ·						3.45 3.60	7.8	
13 0.26	-+	1	118	9.8		ł		9.8 9.8		0.130							3.65		
130.28		<u>+</u>	1 15	3.0				9.3		0.150	ł					·	3.80	8.0	ŧ
12 0.26	+	1	1 13	4.7		-		9.5		0.170							4.00	6.3	ŝ
09 0.27	-	1	1 13	1.9			-	9.9	2.95	0.130								7.8	1
0.42	1			1.8				9.9		0.010								7.8	5
13 0.26		1		0.5				9.9		0.120							2.35	8.00	
13 0.44	0.23	1	2	8.9	0.62	0.2		10.4		0.250	L						3.10	7.80	4
13 0.44	0.24	<b></b>			0.36			10.0		0.200	0.150	L					3 45	7.80	4
16 0.40	0.23	÷	1 15	3.1	0.30	0.2		10.7		0.550		<b></b>				0.15	3.10	8.0	
15 0.42	0.27	+	+ 15	55	0.35	0.5		20.2		0.110			1				3.50	7.9	a-
14 0. 61	10.13		1 12	8.5	0.31	0.2	<u> </u>	10.3		0.130			t					10.50	ŝ
17 0.43	0.22	1	12	3.9	0.09	0.3	-	10.7		0.110							3.10	7.40	đ
								D.	ISCO										
1.1	1	1	12	5.9		[	BA	5.2		1	[	1	1				5.77		T
0.01		1	1 13	1.0				5.2	1								6.00		T
0.01		1	1.13	5.7				5.2									6.00		I
-	1	1	2	5.3				14.3				L	1				6.00		1
0.01	-	1		0.9				14.6	L		· · · ·	L					6.00		+
0.01		<b> </b>		6.3	0.10	ļ	L	14.7	<b> </b>		I	ļ	<u> </u>				6.00		ŧ
0.02	2.30	·	+ 13	2.9	0.10			23.3		h	I	<b> </b>	ŧ				5.35		ŧ
0.04		+		5.9	· · ·	+		24.2	<u> </u>		<u> </u> −−	<b> </b>	+			<u>+  </u>	6.00	h	t
10.01	+	+		5.6		+	۴	14.3		0.290	+	<del> </del>	t			t	6.00		t
on in wel	ght pe	rcent				÷		2.2		10.1230	L	A	ليستعلم		·	d			*
ce																			

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****	r											mt, \$					<u>.</u>	Č4	BE	78	****
Ľ	B	c	<u>A1</u>	<u>\$1</u>	TÍ	V	Cr	Min	P.	Go	Ni	Ĉu	Y	Zr	വ	Mo	La		HT HT	T	·!
· .				-	•			••••••		Nickel Bas	Alloys			·····				·····			T
		0.01-0.04	0.0-9.99	0.0-2.00	0.0-9.00	1	0.0-25.5		0.0-20.0	0.0-10.0			0.0-0.15	0.0-0.50	0.0-7.00	0.0-6.01		0.0-1.0		0.0-12.0	0.
					0.02-3.96		10.1-25.0			0.0-10.0					0.50-0.51	0.1-2.2			1.		1.
			2.65-6.16				1			1						:					1.
		0.11-0.17	2.65-6.20		0.02-3.98		10.1-25.0		1	0.0-24.2					0.50-0.51	•					
	0.006-0 017	0.05-0.10	1.84-3.68		1.80-4.13		18.3-25.0		0.0-0.2	11.8-23.4			0.0-0.200	0.063-0.095	1	0.0-4.1	0.0-1.45			0.0-2.40	0.
	0.006-0.024	0.05-0.14	1.94-4.12		1.30-4.40		13.3.25.0		0.0-0.2	0.0-23.4			0.0-0.200	0.015-0.160	0.0-2.10	0.0-4.1	0.0-1.45			0.0-2.40	0.
, 5, 7-24	0.008-0.024	0.05-0.14	1.94-4.12		1.30-4.40		18.3-25.0	1	0.0-0.2	0.0-23.4			0.0-0.200	0.015-0.160	0.0-2.10	0.0-4.1	0.0-1.45			0.0-2.40	0.
					0.10-4.70	0.0.1.00			0.0-1.7	0.0-25.8				0.0-0.100	0.0-2.37	0.0-6.5		0.0-0.50		0.0-1.90	0
	0.0-0.027	0.05-0.13	1.00-0.50		0.10-4.70	0.0-1.00	3.3-19.4		0.0-1.7	0.0-29.0		-		0.0-0.100	0.0-2.37						
	I	L	L	1		<u> </u>	J	1	1	<u> </u>		l	l	1	I	. I	1	i	4	<u> </u>	- <b>1</b>
		<b>.</b>		·····				·		Cobalt Base	Alloys		· · · · · · · · · · · · · · · · · · ·		· · · · · · · · · · · · · · · · · · ·		T	1	1	F	-F
.64	0.01-0.45				0.0-4.00		0.0-35.0				0.0-10.2		0.0-0.150	0.0-0.500	0.0-4.00	0.0-6.0	0.0-0.15	0.0-1.00		0.0-15.00	0
	0.008-0.015	0.23-0.40	l .			1	27.3-34.7	1	Ì	1	0.5-10.7	0 0-2.95	0.0-0.190		1		1			0.0-4.00	6
		0.26-0.42		1			29.3-34.7			Ş	9.3-9.9		0.0-0.190	[						0.0-4.00	6
.11	0.0-0.015	1		1						1							1			0.0-4.00	6
	0.0-0.014	0.26-0.42		0.0-0.27			29.8-34.7				0.5-9.9		0.0-0.190	[			1			-	6
13	0.0-0.018	0.26-0.61	l				29.5-34.7	0.0-0.62	0.0-0.3		9.3-20.2	0.0-2.95	0.0-0.550	0.0-0.105		1			0.0-0.15	0.0-4 00	
															1				100010	0.0-4.00	
	0.0-0.018	0.23-0.61		0.0-0.27		<u> </u>	27.3-34.7	0.0-0.62	0.0-0.3		0.5-20.2	0.0-2.95	0.0-0.550	0.0-0.105				,	0.0-0.15		6.  -
	Sea Saa Salt							<u> </u>	<u> </u>	E	lement		0.0-0.550	0.0-0.105	1 No 1	La	Ce	 Rf 1 1	C.G0.15	Re	
T	Sea fime Salt		c A		Ti	 		<u> </u>	<u> </u>	E Co Ni	lement Cu		<u> </u>	Cp		La	Ce	Rf 1 1			<u> </u>
1°F  1	Sea salt sour pps	B [	5.0 d +4.	1 <u>Si</u> 78 a -6.		<u> </u>	Cr	 Nn	Fe	E Co Ni Nickel Ba	lement Cu	¥	2r c] -79.6	Cp		La	-11.7 c	<u> </u>	Ta M	Re I cl	·      0.
1675 1750	Sea salt sour pps	B [	5.0 d +4.	1 <u>Si</u> 78 a -6.	33c] -1.73 32c] +0.811	b	-3.66 a	 Nn	Fe	E Co Ni Nickel Ba	lement Cu	-77.3 -77.6	zr 	Cp		Lə		-1.	Ta M	Re 1 c 14c	
1675 1750 1900	Sea Salt Sour ppa 100 200 +1 100 200 +1 100 200 +1 100 200 +1	B B B B B B B B B B B B B B B B B B B	5.0 c +4. 7.0 c +5. 5.0 c +1.	1 51 78 a -6. 50 a -4. 24 c -1.	33c -1.73 2c +0.811 17c -0.72	je	Cr -3.66 a -2.52 a -3.13 a	 Nn	Fe	Zo Ni Nickel Ba 0.492c 0.005c 0.619c	lement Cu	¥	zr 	Cb c -1.69 c dc +1.56 c c +14.5 c	-2.54 c +0.327 c +2.47 c	L>	-11.7 c	-1.	Ta W 17 cl -1.2 059 cl +0.4	1 cl 14c 7 a	
1675 1750 1900 1600 1750	Sea           Saa	B 30.0 c - 266 4.42c +21 52.6 c +69 +61 +22	0 c +4. 0 c +5. 0 c +1.	1 51 78 a -6. 50 a +4. 24 c -1. 6 c - 61 e	33c -1.73 32c +0.811 17c -0.73 -23.0	c	-3.66 a -2.52 a -3.13 a -2.54 c -3.43 b	 Nn	<b>F</b> € -0.707c -0.055c -0.057c + +	E Co N1 N1ckel Ba 0.492c 0.005c 0.619c 6.73 c 0.364c	lement Cu	-77.3 -77.6	zr 	c -1.69 c 3c +1.55 c c +14.5 c -1363.0 c -1362.0 c	-2.54 c +0.327 c +2.47 c -56.7 c		-11.7 c	-1.	17 cl -1.2 059c +0.4 19 cl +3.2 -25.7 +5.2	Re           1 cl           1bC           7 a           c           2 cl	
1675 1750 1900 1600 1750 1500	See           Fine Salt           Sour pps           100 200           100 200           100 200           100 200           500 5           100 200	B 300.0 c] -265 4.42c +21 52.6 c +459 +61 +61 +22 -44	5.0 c +4. 6.0 c +5. 5.0 c +1. 6.0 c -29. 6.0 c -9. 6.6 c -4.	1 51 78 a -6. 50 a +4. 24 c -1. 5 c 51 c 55 c	33c -1.73 32c +0.811 17c -0.73 -23.0 -7.95 -9.25	ус с с	-3.66 a -2.52 a -3.13 a -2.94 c -3.43 b -3.43 a	 Nn	<b>F</b> € -0.707C → -0.055C → -0.057C → + + +	2 Co Nickel Ba 0.492c 0.005c 0.619c 6.73 c 0.864c 0.130c	lement Cu	-77.3 -77.6	zr 	c -1.69 c 3c +1.55 c c +14.5 c -1363.0 c -1362.0 c	-2.54 c +0.327 c +2.47 c -56.7 c	La	-11.7 c	-1.	17         -1.2           059c         +0.4           19         cl           -25.7         +5.2           -3.8         -3.8	Re           1 c1           14c           7 a           c           c           2 c           2 c           2 c           8 c	
1675 1750 1900 1600 1750 1600 1750 1600	See           Fine         Selt           Jour pps         100           100         200           100         200           500         5           100         200           500         5           500         5           100         200           500         5           100         200           500         5           100         5	B 190.0 c - 266 1,82c +21 152.6 c +69 +611 +22 -28 -28	$\begin{array}{c} .0 & c & +4 \\ .0 & c & +5 \\ .0 & c & +1 \\ .0 & c & -29 \\ .0 & c & -9 \\ .6 & c & -4 \\ .0 & c & +420 \\ .0 & c & +1 \\ .0 & c & +1 \end{array}$	1 51 78 a -6, 50 a +4, 24 c -1, 6 c - 6 c - 75 c - 7 a - 79 c	33c -1.73 2c +0.311 17c -0.72 -23.0 -7.95 -9.25 +10.2 -2.50		Cr -3.66 a -2.52 a -3.13 a -2.54 c -3.43 b -4.53 a -4.15 a -4.15 a	<u>Bn</u>	<b>Fe</b> -0.707c -0.055e -0.057e ++ ++ ++	E Co N: Nickel Be 0.092c 0.005c 0.613c 0.613c 0.864c 0.130c 0.130c 0.420c 0.023c	lement Cu	-77.3 -?.6 -34.7	2r c -79.6 7 c -1.1 c -71.4	Cb c −1.55 c 3c +1.55 c c +14.5 -1363.0 c -1922.0 c -931.0 +2134.0	-2.54 c +0.327 c +2.17 c -56.7 c +21.6 c +21.6 c +10.7 c +10.7 c		-11.7 c -1.07c -12.7 c		If         c         -1.2           0559c         +0.4         -1.9           19         c         +3.2           -25.7         +5.2           -3.0         -3.4           +0.3         +0.3	1 cl 14c 14c 7 a 2 cl 2 cl 2 cl 2 cl 8 c 66c	
1675 1750 1900 1600 1750 1600 1750 1600 1750 1750 1750	Sea ime §5alt           1001 200         +1           100200         +3           100200         +3           100200         +3           100200         +3           100200         +3           100200         +3           100200         5           100200         5           100200         5           100200         5           100200         5           100200         5           100200         5	B 4.420 +21 52.6 c +69 +61 +22 -44 -67 -67 -28 -69,0 c -11	5.0 c +4. 6.0 c +5. 5.0 c +1. 5.0 c -29. 5.0 c -29. 6.0 c -29. 7.6 c -4. 7.0 s +20. 7.0 c +12.	1 51 78 a -6. 24 c -1. 24 c -1. 6 c 61 c 25 c 25 c 29 a 79 c 9 c	33c -1.73 2cc +0.811 17c -0.79 -23.0 -7.95 -9.25 +10.2 -2.50 +18.2 +34.2		Cr -3.66 a -2.52 a -3.13 a -2.54 c -3.43 b -4.15 a -4.15 a -4.15 a -4.15 c -3.96 a +5.42 c	Hn	9e           -0.707c           -0.055c           -0.057c           -0.057c           +      <	E Co N1 N1ckel Bs 0.492c 0.005c 0.619c 6.73 c 0.130c 0.420c 0.420c 0.023c 1.79 a 5.4 c	lement Cu	-77.3 -7.6 -34.7 	2x c -73.6 c -1.1 c -71.4 b +538.0 c +1563.5	Cb C −1.69 · C +1.55 · C +11.55 · C +11.5 · -1363.0 · -236.0 · -231.0 · -	-2.54 c +0.327 c +2.47 c +2.47 c +2.16 c +3.44 c +10.7 c +17.5c c -3.27 b -6.37 c	-9.33 c -24 1 c	-11.7 c -1.07c -12.7 c	-1. +0. +1.	17 $-1.2$ $059c$ $+0.8$ $19$ $c$ $-25.7$ $+5.2$ $-3.0$ $-3.8$ $-7.2$ $-7.2$ $-3.2$ $-3.0$ $-3.3$ $-99.9$	Re           1 c           14C           7 a           2 c           2 c           2 c           2 c           66c1           c           c	0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0
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*7 1 1675 1750 1300 1600 1600 1750 1600 1750 1750 1750 1750 1750 1600 1750 1600 1750 1600 1750 1600 1750 1600 1750 1600 1750 1600 1750 1600 1750 1600 1750	Sea           Salt           Sour         span           100         200         +)           100         200         +)           100         200         +)           100         200         +)           100         200         -)           500         5         -)           100         200         -)           500         5         -)           100         200         ->           500         5         -)           1000         5         ->           1000         5         ->           1000         5         ->           1000         5         ->           1000         5         ->           1000         5         ->           1000         5         ->           10000         5         ->           1000         5         ->           1000         5         ->           1000         5         ->           1000         5         ->           1000         5         ->           1000         5	B 30.0 C 286 4.82C 421 152.6 C 459 461 422 444 -25 -25 -25 -25 -25 -25 -25 -25	5.0 c +4. 7.0 c +5. 5.0 c +1. 5.0 c -29. 1.6 c -4. 5.0 c +12. 5.0 c +12. 5.2 c +22. 5.2 c +22. 5.2 c +22. 5.2 c +20. 5.2 c +20. 5.5 c +70.	1 51 78 a -6. 24 c -1. 5 c - 51 e - 25 c - 9 a - 79 c - 9 c - 70 v -	326 -1.73 266 +0.811 17c -0.72 -23.0 -23.0 -23.0 -2.50 +10.2 -2.50 +18.2 -0.30 +18.2 -0.50 +0.771 -6.66	5e c c c c c c c c c c c c c c c c c c c	Cr -3.66 a -2.52 a -2.13 a -2.94 c -3.43 b -4.55 a -4.15 a -4.55 a -4.15 a -4.55 a -4.75 a	Rn	Pe           -0.707c           -0.055c           -0.057c	E           CO         Ba           Nackel Ba         Ba           0.492c         0.005c           0.0512         0.005c           0.492c         0.0025c           0.420c         0.0025c           0.420c         0.0025c           0.110         0.025c           0.025c         0.025c           0.053c         0.055c           0.055c         0.055c           0.055c         0.055c           0.055c         0.055c           0.055c         0.055c	lement Cu se Alloys	-77.3 -7.6 -34.7 +52.1 +52.1 -2.6 -2.6 +5.0	2x c -73.6 c -1.1 c -71.4 b +530.0 c +1550.5 1 c +20.2 3 c +2.5 5 c +2.5 -1140.3 a1 -54.4	Cb           c         -1.65           3c         +18.5           c         +18.5           -1363.0         -1922.0           -931.0         +2134.0           +2134.0         +2134.0           c         c           c         -0.332.0           d         -1.073.2           a         -103.0	-2.54 c +0.327 c +2.16 c +2.16 c +2.16 c +2.16 c +10.7 c +17.5c c +17.5c c -3.27 c -6.37 c +17.5c c +1.54 c +1.1 <sup>1</sup> c +1.1 <sup>1</sup> c +1.1 <sup>1</sup> c	-3.33 c -24 1 c +0.210c +0.352c -0.724 c	-11.7 c -1.07c -12.7 c +3.46c -27.1 a	-1. +0. +1. +1. +1. +2. -0. +0. +0. +56	$17^{\circ}$ cl $-1.2$ $059c$ $+0.R$ $19^{\circ}$ cl $+3.2$ $-3.0$ $-3.0$ $-7^{\circ}$ cl $-12.3$ $7^{\circ}$ cl $-12.3$ $7^{\circ}$ cl $-12.3$ $7^{\circ}$ cl $-12.3$ $7^{\circ}$ cl $-12.3$ $5^{\circ}$ cl $-0.3$ $5^{\circ}$ cl $-0.3$	Re           1 cl           14c:           7 a           c           2 cl           2 cl           2 cl           8 cl           66c1           c           c           3 cl           3 cl           1 LL           72 cl           3 cl           1 LL           72 cl           1 LL           1 Cc	
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1675 1750 1500 1750 1600 1750 1600 1750 1600 1750 1750 1600 1750 1750 1750 1750 1600 1750 1750 1750 1750 1600 1750 1750 1750 1750 1750 1750 1750 17	Sea           Size           Salt           Sour         span           100         200           100         200           100         200           100         200           100         200           500         5           100         200           500         5           100         200           500         5           100         200           500         5           100         200           500         5           1000         5           5000         5           1000         5           1000         5           5000         5           1000         5           1000         5           500         5           100         200           100         200	B 4,820 + 21 4,820 + 21 152.6 C + 659 + 611 + 222 - 67 - 67 - 29 - 20 - 20	5.0 c +4. 7.0 c +5. 5.0 c +1. 5.0 c -2. 5.0 c -2. 5.0 c -2. 5.0 a 420. 5.0 a 420. 5.0 c +12. 5.2 c +32. 5.2 c +32. 5.2 c +32. 5.4 c -0. 5.5 c +0. 5.5 c +0. 5.5 c +70. 7.3 b 7.4 c -1. 5.0 c -2. 5.5 c -2. 5.5 c +0. 5.5 c +0.	1 51 78 a -6. 24 c -1. 5 c - 51 e - 25 c - 9 a - 79 c - 9 c - 70 v -	326 -1.73 266 +0.811 17c -0.72 -23.0 -23.0 -23.0 -2.50 +10.2 -2.50 +18.2 -0.30 +18.2 -0.50 +0.771 -6.66	c c c c c c c c c c c c c c c c c c c	Cr -3.66 a -2.52 a -2.52 a -3.13 a -2.54 a -3.15 a -3.26 a +5.42 c -4.15 a -3.26 a +5.42 c -0.437 c -0.457 c -4.65 a -4.65 a -4.65 a -4.65 a -2.59 a -2.59 a -3.65 a -3.26 a -4.65 a -4.65 a -4.65 a -4.65 a -4.65 a -2.36 a -3.26 a -2.30 a -2.30 a	Rn	Pe           -0.707c           -0.055c           -0.057c	B           CO         Ba           Nackel Ba         0.492c           0.055c         0.055c           0.0513c         0.613c           0.130c         0.420c           0.420c         0.023c           1.79         5.4           0.035c         0.053c           0.053c         0.053c           0.053c         0.053c           0.053c         0.053c           0.053c         0.70           Cobalt Ba         -0.1           0.02         -0.2	lement Cu se Alloys se Alloys se Alloys 500c 68c 2835c	-77.3 -7.6 -84.7 -84.7 -152.1 +52.1 +52.1 +52.3 -6.0 -2.6 +5.0 -178.0 -178.0 -127.0	$ \begin{array}{c}       2r \\       c & -79.6 \\       c & -1.1 \\       c & -71.4 \\       c & -71.4 \\       c & -71.5 \\       c & -11559.5 \\       c & +1559.5 \\       c & +1559.5 \\       c & +2.5 \\       c & -1140.2 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\   \end{array} $	Cb           c         -1.69           3c         +1.55           c         +14.5           -1363.0         -1922.0           -1922.0         -1923.0           -2931.0         +2134.0           +2134.0         -2434.0           -243.0         -2433.0           c         +1.27           -1033.0         -1033.0           a         -16.72           a         -6.72           a         -6.55	-2.54 c +0.327 c +2.16 c +2.16 c +2.16 c +2.16 c +10.7 c +17.5c c +17.5c c -3.27 c -6.37 c +17.5c c +1.54 c +1.1 <sup>1</sup> c +1.1 <sup>1</sup> c +1.1 <sup>1</sup> c		-11.7 c) -1.07c -12.7 c -12.7 c +3.46c +3.46c	-1. +40. +11. +11. +20. +0. +0. +0. +0. +0. +0. +0. +0. +0. +	Na         W           117 cl         -1.2           059 cl         +0.8           19 cl         +3.2           -25.7         +5.2           -3.6         -3.6           -7 cl         -12.5           3 cl         -36.2           -7 cl         -12.5           -3.6         +0.6           -5 cl         -0.3           -613 cl         -0.3           5 cl         -0.3           5 cl         -0.3           -5 cl         -0.3           -603 al         -2.0           -37 cl         +1.4           -605 cl         +1.4	Re           1 c           14c           7 a           7 a           2 c           2 c           2 c           3 c           57c           97c           10 c           97c           10 c           97c           10 c           10 c           11 a           11 c	
*r 1 1675 1750 1500 1750	Sea           Size         \$a1t           Jour ppa         100           100         200           100         200           100         200           100         200           100         200           500         5           100         200           500         5           100         200           1000         5           10000         5           5000         5           10000         5           10000         5           100         200           100         200           100         200           100         200           100         200           100         200	B 90.0 c) -26 4,32c +21 152.6 c +459 152.6 c +459 461 +451 152.6 c -22 249.0 c -11 153.7 c +11 159.9 c -11 159.9 c -11 159.9 c -11 159.9 c +14 159.9 c +14 159.0 c +14 159.	5.0 c 44. 7.0 c 45. 5.0 c 41. 5.0 c 41. 5.0 c -2. 5.0 c -4. 5.0 c -4. 5.0 c 41. 5.0 c -4. 5.0 c 41. 5.2 c +22. 5.2 c +20. 5.5 c +0. 5.5 c +0. 5.0 c +1. 5.0 c +1. 5	1 51 78 a -6. 24 c -1. 5 c - 51 e - 25 c - 9 a - 79 c - 9 c - 70 v -	33c -1.73 7c -0.73 -23.0 -7.95 -25.0 -10.2 -2.50 +10.2 -2.50 +11.2 -2.50 +11.2 -2.50 -2.50 +11.2 -2.50 -2	c c c c c c c c c c c c c c c c c c c	Cr -3.66 a -2.52 a -2.52 a -3.13 a -2.54 d -3.26 a -3.43 b -3.26 a -3.26 a -1.15 a -1.16 a -1.16 a -2.30 a +0.213 c +0.213 c +0.213 c -0.413 c -1.16 a -2.30 a +0.213 c -0.413 c -1.16 a -2.30 a -4.05 a -5.05 a -5	Rn	Pe           -0.707c           -0.055c           -0.057c	B           CO         Ba           Nackel Ba         0.492c           0.055c         0.055c           0.0513c         0.613c           0.130c         0.420c           0.420c         0.023c           1.79         5.4           0.035c         0.053c           0.053c         0.053c           0.053c         0.053c           0.053c         0.053c           0.053c         0.70           Cobalt Ba         -0.1           0.02         -0.2	lement Cu se Alloys se Alloys se Alloys 500c 68c 2835c	-77.3 -7.6 -84.7 -84.7 -152.1 +52.1 +52.1 +52.3 -6.0 -2.6 +5.0 -178.0 -178.0 -127.0	$ \begin{array}{c}       2r \\       c & -79.6 \\       c & -1.1 \\       c & -71.4 \\       c & -71.4 \\       c & -71.5 \\       c & -11559.5 \\       c & +1559.5 \\       c & +1559.5 \\       c & +2.5 \\       c & -1140.2 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\   \end{array} $	Cb           c         -1.69           3c         +1.55           c         +14.5           -1363.0         -1922.0           -1922.0         -1923.0           -2931.0         +2134.0           -42134.0         -2424.0           -1035.4         -1033.0           a         -103.0           a         -6.72           a         -6.55	-2.54 = +0.327 c +2.17 c +2.17 c +2.17 c +2.17 c +2.16 c +2.16 c +2.16 c +2.16 c +2.16 c +2.16 c +2.17 c +10.7 c +1.17 c -3.27 b -6.37 c +1.12 c +1.12 c +1.12 c +1.12 c +1.12 c +1.12 c +1.01 c +1.01 c +1.01 c		-11.7 c) -1.07c -12.7 c -12.7 c +3.46c +3.46c	-1. +4C. +11. +11. +11. +11. +11. +29. +00. +00. +00. +00. +00. +00. +00. +0	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	
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1675 1675 1750 1500 1750	Sea           Time Salt           Sour ppa           100 200         +1           100 200         +0           100 200         +0           100 200         5           100 200         5           100 200         5           100 200         5           100 200         5           100 200         5           100 200         5           100 200         5           100 200         5           100 200         5           100 200         100           100 200         100           100 200         5           100 200         100           100 200         100           100 200         5           100 200         100           100 200         100           100 200         5           100 200         5           100 5         4           1000 5         4           1000 5         4           1000 50         5	B 30.0 c - 266 4.52c 421 152.6 c +59 +61 +629 +629 -67 -67 -75 -25 -25 -25 -25 -25 -25 -25 -2	5.0 $c$ 44. 7.0 $c$ 45. 6.0 $c$ -29. 7.0 $c$ -29. 7.0 $c$ -29. 7.0 $c$ -41. 7.0 $c$ -41. 7.0 $c$ -41. 7.2 $c$ -41. 7.2 $c$ +12. 7.2 $c$ +12. 7.2 $c$ +12. 7.3 $c$ -0. 7.4 $c$ -0. 7.5 $c$ +70. 7.3 $b$ -7. 7.4 $c$ -3. 7.4 $c$ -3. 7.5 $c$ +70. 7.5 $c$ +70. 7.5 $c$ +70. 7.5 $c$ -40. 7.5 $c$ +70. 7.5 $c$ -40. 7.5 $c$ -70. 7.6 $c$ -70. 7.7 $b$ -70. 7.7 $b$ -70. 7.8 $c$ -70. 7.9 $b$ -70. 7.9 $b$ -70. 7.9 $c$	1 51 78 a -6. 24 c -1. 5 c - 51 e - 25 c - 9 a - 79 c - 9 c - 70 v -	33c -1.73 7c -0.73 -23.0 -7.95 -25.0 -10.2 -2.50 +10.2 -2.50 +11.2 -2.50 +11.2 -2.50 -2.50 +11.2 -2.50 -2	e c c c c c c c c c c c c c c c c c c c	$\begin{array}{c c} cr \\ \hline -2.52 a \\ \hline -2.52 a \\ \hline -2.52 a \\ \hline -2.13 a \\ \hline -2.53 a \\ \hline -3.13 a \\ \hline -3.15 a \\ \hline -4.15 a \\ \hline -3.96 a \\ \hline +5.82 c \\ \hline -0.055 c \\ \hline -0.055 c \\ \hline -4.73 c \\ \hline -0.055 c \\ \hline -4.73 c \\ \hline -0.055 c \\ \hline -4.73 c \\ \hline -0.055 c \\ \hline -4.65 a \\ \hline -0.055 c \\ \hline -4.73 c \\ \hline -0.055 c \\ \hline -0$	Rn	Pe           0.707c           -0.055c           -0.057c	B           CO         Ba           Nackel Ba         0.492c           0.055c         0.055c           0.0513c         0.613c           0.130c         0.420c           0.420c         0.023c           1.79         5.4           0.035c         0.053c           0.053c         0.053c           0.053c         0.053c           0.053c         0.053c           0.053c         0.70           Cobalt Ba         -0.1           0.02         -0.2	lement Cu se Alloys se Alloys se Alloys 500c 68c 2835c	-77.3 -7.6 -84.7 -84.7 -152.1 +52.1 +52.1 +52.3 -6.0 -2.6 +5.0 -178.0 -178.0 -127.0	$ \begin{array}{c}       2r \\       c & -79.6 \\       c & -1.1 \\       c & -71.4 \\       c & -71.4 \\       c & -71.5 \\       c & -11559.5 \\       c & +1559.5 \\       c & +1559.5 \\       c & +2.5 \\       c & -1140.2 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\   \end{array} $	Cb           c         -1.69           3c         +1.55           c         +14.5           -1363.0         -1922.0           -1922.0         -1923.0           -2931.0         +2134.0           -42134.0         -2424.0           -1035.4         -1033.0           a         -103.0           a         -6.72           a         -6.55	-2.54 = +0.327 c +2.17 c +2.17 c +2.17 c +2.17 c +2.16 c +2.16 c +2.16 c +2.16 c +2.16 c +2.16 c +2.17 c +10.7 c +1.17 c -3.27 b -6.37 c +1.12 c +1.12 c +1.12 c +1.12 c +1.12 c +1.12 c +1.01 c +1.01 c +1.01 c		-11.7 c) -1.07c -12.7 c -12.7 c +3.46c +3.46c	-1. +40. +11. +11. +24. +0. +0. +0. +0. +0. +0. +0. +0. +0. +0	$17^{\circ}$ cl $-1.2$ $059^{\circ}$ cl $40.8$ $19^{\circ}$ cl $+3.2$ $-3.6$ $-5.8$ $-9.5, 7$ $+5.2$ $-3.6$ $-5.8$ $-0.7$ $-12.5$ $3^{\circ}$ cl $-9.0$ $-712c^{\circ}$ cl $-0.3$ $-5c^{\circ}$ cl $-0.3$ $-0.37c^{\circ}$ cl $+1.2$ $-0.75^{\circ}$ cl $-1.90^{\circ}$ $-0.75^{\circ}$ cl $-1.90^{\circ}$ $-0.75^{\circ}$ cl $-0.20^{\circ}$ $-0.75^{\circ}$ cl	$ \begin{array}{c c}                                    $	
*r 1 1675 1750 1600 1750 1750 1750 1750 1750 1750 1750 17	Sea           Size           Salt           Sour         pp           100         200           100         200           100         200           500         5           100         200           500         5           100         200           500         5           100         200           500         5           100         200           500         5           100         200           500         5           1000         5           1000         5           1000         5           1000         200           1000         5           1000         200           1000         200           1000         5           1000         5           5000         5           1000         200           1000         5           2000         5           1000         5           2000         5	B 90.0 c - 266 4,32c + 21 152.6 c + 459 + 611 + 451 - 65 - 22 153.7 c + 11 - 59.3 c - 11 - 59.3 c - 12 - 108.0 c - 49 - 44 - 45 - 4	5.0 c +4. 7.0 c +5. 5.0 c +1. 5.0 c +1. 5.0 c -2. 5.0 c -4. 5.0 c +1. 5.0 c +1. 5.0 c +1. 5.0 c +1. 5.2 c +2. 5.2 c +2. 5.2 c +2. 5.5 c +0. 5.5 c +0.	1 51 78 a -6. 24 c -1. 5 c - 51 e - 25 c - 9 a - 79 c - 9 c - 70 v -	33c -1.73 7c -0.73 -23.0 -7.95 -25.0 -10.2 -2.50 +10.2 -2.50 +11.2 -2.50 +11.2 -2.50 -2.50 +11.2 -2.50 -2	e c c c c c c c c c c c c c c c c c c c	Cr -3.66 a -2.52 a -3.13 a -2.54 c -3.43 b -4.53 a -4.53 a -4.53 a -4.53 a -4.53 a -4.65 a -0.437c -0.437c -0.055c -4.73 c -1.65 a -2.30 a -2.30 a -4.63 a -2.30 a -2.30 a -2.30 a -4.63 a -2.30 a -2.30 a -4.63 a -2.30 a -2.30 a -2.30 a -2.30 a -2.230 a -0.025c -0.025c -2.230 a -0.025c -2.230 a -0.025c -2.230 a -0.025c -2.230 a -0.025c -2.230 a -0.025c -2.300 a -0.025c -2.500 a -2.500 a -2.5000 a -2.5000 a -2.5000 a -2.5000 a -2.5000 a -2.5000 a -2	Rn	Pe           0.707c           -0.055c           -0.057c	B           CO         N3           Nickel Be         0.492c           0.052c         0.055c           0.055c         0.055c           0.056c         0.0366c           0.022c         0.130c           0.420c         0.025c           0.025c         0.025c           0.036c         0.036c           0.036c         0.036c           0.056c         0.036c           0.056c         0.0420c           0.056c         0.036c           0.056c         0.0420c           0.056c         0.036c           0.056c         0.0420c           0.0420c         0.0420c           0.056c         0.0420c           0.056c         0.0420c           0.056c         0.0420c           0.056c         0.0420c           0.056c         0.0420c           0.057c         0.0400c           0.050c         0.040	1000001 1000001 1000001 100001 100001 100001 1000000 1000000 1000000 100000000	-130.0 -177.3 -7.6 -34.7 +52.1 +95.3 -6.0 -9.6.6 -9.6.7 -9.6.7 -9.6.7 -9.7 -9.7 -9.7 -9.7 -9.7 -9.7 -9.7 -9	$ \begin{array}{c}       2r \\       c & -79.6 \\       c & -1.1 \\       c & -71.4 \\       c & -71.4 \\       c & -71.5 \\       c & -11559.5 \\       c & +1559.5 \\       c & +1559.5 \\       c & +2.5 \\       c & -1140.2 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\   \end{array} $	Cb           c         -1.69           3c         +1.55           c         +14.5           -1363.0         -1922.0           -1922.0         -1923.0           -2931.0         +2134.0           -42134.0         -2424.0           -1035.4         -1033.0           a         -103.0           a         -6.72           a         -6.55	-2.54 = +0.327 c +2.17 c +2.17 c +2.17 c +2.17 c +2.16 c +2.16 c +2.16 c +2.16 c +2.16 c +2.16 c +2.17 c +10.7 c +1.17 c -3.27 b -6.37 c +1.12 c +1.12 c +1.12 c +1.12 c +1.12 c +1.12 c +1.01 c +1.01 c +1.01 c		-11.7 c) -1.07c -12.7 c -12.7 c +3.46c +3.46c	-1. +40. +11. +11. +11. +11. +11. +29. +0. +0. +0. +0. +0. +0. +0. +0. +0. +0	IT         cl         -1.2           059c         +0.4         -1.9           19         cl         +5.2           19         cl         +5.2           -3.0         -3.6           -3.7         -12.5           -3.6         -3.6           -3.7         -12.5           -3.6         -3.6           -3.7         -12.5           -3.6         -0.2           -5.5         -0.3           -5.5         -0.3           -5.5         -0.3           -5.7         -1.2           -0.8         -2.2           -10         al           -1.2         -0.6           -5.5         -0.3           -5.5         -0.3           -5.7         -0.2           -10.8         -2.2           -10.8         -2.2           -1.2         -0.2           -1.2         -0.2           -1.2         -1.2           -1.2         -1.2           -1.2         -1.2           -1.2         -1.2           -1.2         -1.2           -1.2         -1.2	Re           1 c             14c           7 a             7 a             2 c             2 c             2 c             2 c             2 c             3 c             2 c             2 c             2 c             2 c             2 c             2 c             3 c             1 a             1 a             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             3 a             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c             1 c	
* 7 1 1675 1750 1600 1600 1750 1600 1750 1500	See           Size           Size           Sourt ppe	B 90.0 c - 266 4,32c + 21 152.6 c + 459 + 611 + 451 - 65 - 22 153.7 c + 11 - 59.3 c - 11 - 59.3 c - 12 - 108.0 c - 49 - 44 - 45 - 4	5.0 $c$ +4. 7.0 $c$ +5. 5.0 $c$ +1. 5.0 $c$ -2. 5.0 $c$ +1. 5.0 $c$ -2. 5.0 $a$ +20. 5.0 $a$ +20. 5.0 $c$ +12. 5.0 $c$	1 51 78 a -6. 24 c -1. 5 c - 51 e - 25 c - 9 a - 79 c - 9 c - 70 v -	33c -1.73 7c -0.73 -23.0 -7.95 -25.0 -7.95 -9.25 -9.25 -10.2 -2.50 +10.2 -2.50 +11.2 -2.50 +13.2 +34.2 -0.505	c c c c c c c c c c c c c c c c c c c	Cr -3.66 a -2.52 a -2.52 a -2.53 c -3.13 a -2.54 c -3.43 b -4.15 a -4.15 a -4.15 a -4.55 a -4.65 a -2.30 a -0.942b -0.942b -0.942b -0.942b -0.942b -0.922c	Rn	Pe           0.707c           -0.055c           -0.057c	B           CO         N3           Nickel Be         0.492c           0.052c         0.055c           0.055c         0.055c           0.056c         0.0366c           0.022c         0.130c           0.420c         0.025c           0.025c         0.025c           0.036c         0.036c           0.036c         0.036c           0.056c         0.036c           0.056c         0.0420c           0.056c         0.036c           0.056c         0.0420c           0.056c         0.036c           0.056c         0.0420c           0.0420c         0.0420c           0.056c         0.0420c           0.056c         0.0420c           0.056c         0.0420c           0.056c         0.0420c           0.056c         0.0420c           0.057c         0.0400c           0.050c         0.040	1000001 1000001 1000001 100001 100001 100001 1000000 1000000 1000000 100000000	-77.3 -7.6 -84.7 -84.7 -152.1 +52.1 +52.1 +52.3 -6.0 -2.6 +5.0 -178.0 -178.0 -127.0	$ \begin{array}{c}       2r \\       c & -79.6 \\       c & -1.1 \\       c & -71.4 \\       c & -71.4 \\       c & -71.5 \\       c & -11559.5 \\       c & +1559.5 \\       c & +1559.5 \\       c & +2.5 \\       c & -1140.2 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\       c & -37.5 \\   \end{array} $	Cb           c         -1.69           3c         +1.55           c         +14.5           -1363.0         -1922.0           -1922.0         -1923.0           -2931.0         +2134.0           -42134.0         -2424.0           -1035.4         -1033.0           a         -103.0           a         -6.72           a         -6.55	-2.54 = +0.327 c +2.17 c +2.17 c +2.17 c +2.17 c +2.16 c +2.16 c +2.16 c +2.16 c +2.16 c +2.16 c +2.17 c +10.7 c +1.17 c -3.27 b -6.37 c +1.12 c +1.12 c +1.12 c +1.12 c +1.12 c +1.12 c +1.01 c +1.01 c +1.01 c		-11.7 c) -1.07c -12.7 c -12.7 c +3.46c +3.46c	-1. +40. +11. +20. +0. +0. +0. +0. +0. +0. +0. +0. +0. +	Na         W           117 cl         -1.2           059 cl         +0.8           119 cl         +3.2           -3.6         -3.6           -7         -12.2           -3         -12.7           -3.6         +0.8           -7         -12.7           3         -19.9           -613e         +0.6           -5         c           108         -1.2           063         -1.2           072         +1.2           077         c           10         al           -1.2         -0.3           5         c           077         c           077         c           077         c           077         c           031         z           04//cc         +0.2	Re           1 c           14C           7 a           1 a           1 a           1 a           1 a           1 a           1 a           1 a           1 a           1 a           1 a           1 a           1 a           1 a           1 a      <	

fidence level, b - 90% confidence level, c - Confidence level is equal to or greater than 60% but less than 90% ture, R - Multiple correlation coefficient, S<sub>g</sub> - Standard error of estimate, N - Number of tests per regression analysis, I-Intercept value for regression equation

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e e		and the second s	1 ( ) - 4 <b>3</b>	· · · · · · · · · · · · · · · · · · ·		5	and the second

<u>.</u>		mt 4	A state						<u> </u>			وببيجد
,	Ni	Ċμ	Ý	Zr	Co.	No	La	C+	HE	Ta	· •	<u>Ré</u>
l Bas	e Alloys										·	·
0.0			0.0-0.15	0.0-0.50	0.0-7.00	0.0-6.01		0.0-1.0		0.0-12.0	0.0-24.0	
2.0				,	0.50-0.51	0.1-2.2					1.77-7.49	
4.5					0.50-0.51	0.1-2.2					1.77-7.49	
23.4			0.0-0.200	0.063-0.095		0.0-4.1	0.0-1.45			0.0-2.40	0.0-6.10	0.0-4.5
3.4			0.0-0.200	0.015-0.160	0.0-2.10	0.0-4.1	0.0-1.45			0.0-2.40	0.0-6.10	0.0-4.5
5.4			0.0-0.200	0.015-0.160	0.0-2.10	0.0-4.1	0.0-1.45			0.0-2.40	0.0-6.10	0.0-4.5
5.8				0.0+0.100	0.0-2.37	0.0-6.5		0.0-0.50		0.0-1,90	0.0-12.3	
Base	Alloys	l	İ		<u> </u>	L		l <u>+</u>	l		÷	I
	0.0-10.2		0.0-0.150	0.0-0.500	0.0-4.00	0.0-6.0	0.0-0.15	0.0-1.00		0.0-15.00	0.0-15.00	
	0.5-10.7	0.0-2.95	0.0-0.190			· .				0.0-4.00	6.30-8.10	
	9.3-9.9	0.0-2.95	0.0-0.130							0.0-4.00	6.30-8.25	
	0.5-9.9	0.0-2.95	0.0-0.190							0.0-4.00	6.30-8.25	
	9.3-20.2	0.0-2.95	0.0-0.550	0.0-0.105	· · ·	· .			0.0-0.15	0.0-4.00	6.30-10.50	
	0 5-20 2	0.0-2.95	0.0-0.550	0.0-0.105	1 ·		ł	I	0.0-0.15	0.0-4.00	6.30-10.50	1

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Eles	perit.	•													
Ni	Cu	Y.	Zr	Cb.	Ho	La	Ce .	Ħf	1 7.4		Re	<u> </u>	5	<u> </u>	
1 Base	Alloys	,						-							
		-77.3 c	-79.6 c	-1.69	d -2,54 cl		-11.7 c		-1.17 c	-1.21 c		10.56	26.9	1111	72.
	+	-2.69 c	the state of the second states		C +0.827 C		-1.07c		+0.059c	+0.4140		0.53	25,6	111	2
	-	-54.7 c	-71.4 C		a +2.47 c		-12.7 C		+1.19 c	+3.27 *		10.67	26.1	108	2
		1			1							1	1		
	· +			-1863.0	c -56.7 c				1	-25.7c ¢		0.56	22.3	12	
				-1922.0	c+21.6 c				T.	+5.22 c		0.85	12.9	20	994.
					ci +9.44 c				T	-3.02 c		0.53	16.2	16	161.
				-931.0	d+10.7 d				1	-3.88 c		0.85	20.2	23	533.
	+	· · · ·			d+17.5c c				1	+0.366c		0.91	114.1	12	-1003.
		+52.1 b	+538.0 c		-3.27 b	-9.33 c			+11.7 c	-12.3 c		10.93	0.67	1 31	-251.
			+1569.5 c		-6.97 c	-24.1 ¢			+29.8 c	-99.9 c		0.93	2.44	13	-267.
		-6.01 =	+20.2 c	+1.27	-0.420 c	+0.210¢			-0.712c	-0.087c	-0.3040	0.88	0.61	1 46	18.
	+	-2.63 c	+2.55c			+0.352c			+0.336c	+1.08 b	+1.44 0	0.99	0.59	42	-10.
		+5.05 c		+0.632	c +1.14 c	-0.724 c			+0.619.	+0.972c	+1.35	0.96	0.50	25	-5.
			-1140.0 a		d +2.25 d		+3.46c		1-56.5 c	-0.316c		10.39	26.3	42	-91.

ilt Base Alloys							
-0.600cl  -	130.0 a -5	4.4 a -6.72 a	-1.01 c -2713.9	a -27.1 a	-2.10 a -1.98 a	0.83 14.3	43 128.0
-0.563c	178.0 a -5	4.0 a -6.65 a	+4.07 b-2676.0	a -27.2 a	-2.08 a -2.01 a	0.36 16.2	48 128.0
+0.035c	127.0 3	7.5 cl -4.04 cl	+21.5 a -1883.0	c - 19.1 c	-1.37 c +1.21 c	0.82 24.2	49 74.1
+0.544c -2.12 c	-0.323c				-1.66 c +1.41 c	0.92 1.69	11 -18.9
-0.207c +0.053c	-33.0 a				+0.437 c +0.965 c	0.35 1.43	15 1.9
+0.030c -0.177c	+5.01 d				+0.077c +0.001c	0.96 0.290	11 2.7
+1.30 al -3.55 al	+25.C b			1	-2.81 a -2.83 a	0.82 2.60	28 47.9
-0.009c +0.027c	+0.350b				+0.040c +0.045c	0.99 0.031	11 1.0
+0.294c +0.003c	+0.647 c				+0.047c +0.091c	1.0 0.013	8 -4.4
-6.68 c +0.041c -	31.9 c	1			+0.242c -0.255c	0.90 1.73	16 +75.9
-54.1 cl-16.3 cl+	46.1 c				-13.9 c -17.4 c	0.996 2.62	8 +1073.0
-0.183c -0.133c -	40.4 a				+0.963c +1.52 a	0.95 1.04	20 -43.2
+21.7 cl-20.4 b	21.4 c +57	9.0 c		-197.0c	+10.0 c -3.38 c	0.91 5.22	22 - 358.0
+0.771al -2.14 cl	+0.435 c +2	5.1 cl	1	-19.6c	-1.61 b -0.124c	0.63 2.93	42 +15.7

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#### ysis, I - Intercept value for regression equation

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#### Table 5-A

# Ranges for Each Alloying Element

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#### Table 6-A

# Regression Coefficients for Surface Loss

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#### Table 7-A

#### Regression Coefficients for Maximum Penetration

	r	Time	Sea												-								ement			<b>.</b>			M	<u> </u>	La		Ce	Hf	Ta
		hour			Т	C	- 1	A		Si	T	Tì	T	V	Cr	-	Mn.	Fe		Co		Ni		Cu	Y	Zr	Ch	)	1910	<u> </u>	1.4			1 11	1
			1.55	·																															
																						ase All	oys					Cha	7 7	00 1		1+3		2	1 -0.9
	1675	100	1 200	-271.0	D d	-117.0	0 cl	+3.5	7 1	-1.2	4c	-2.01	a		-3.6					-0.87					+76.5			.54c	+0.1	09 d			.4	3	-0.0
	1750	100	200							+1.6	9c	+0.30	9c		-3.9					-0.62					+170.0 0	+267.0		.900	+8.	<u>2774</u>			3.6	2	13.0
	1900		200							+19.1	b	-1.47	b		-3.1			-0.	055c	-0.67	<u>0 c</u>		_		+38.5	+127.0			+26.			172.		<u> </u>	+
			200			+345.0						-7.15	C		-2.2					-0.00			_						-8.					+	+
			200			+330.					1-	27.1	d		-8.1					+3.95			_				1-2056	.0 c	+78.	<u>22-q</u>					+
	1600	500	5		-	-15.						-2.02	e		1 -5.7					-0.26			_			.j	-2524	2.0 C	+17.	<del>7 </del>					· · · · ·
			200		-	.448.					Τ-	13.1	c		-9.8					+0.44							-151	<u></u>	TIC	<u> </u>				+	i
		500				-597.					-	10.2	Ъ		-6.0					-0.80							- 192	<u>, 0 c</u>	+26.	<u>0</u>				+	+
		1100						+1.			T	+0.43	7 c		-4.7	0 a,				+0.00						1	+305	1.0 c	+ <u>++</u> (·	55 c	-5.15	<u> </u>		- <del> </del>	† +9.
		100	200	+1583.	0 d						1	17.2	d		+5.4	7 c		- 39.		+2.97						+705.0	<u>q</u>		-14.		-51.7				1+52.
	1750			+15084.	i d	- 302.	q c	+78.4	5 d		1	72.2	d		+31.8			- 7839.		+8.05			_			+2731.3					-2.36				1 -2.
		1000		-1023.								-3.62	d		+1.1	4 c				+1.22						+151.0		1.55c		47 c	-0.96			-+	1 -0.
7-24	1600			-512.							_	-1.25	d		+3.1	4 c				+0.78			_		-54.3			5.35c			-0.31			- <del>1</del>	1 -0.
1		1000		+250.		-5.	20c	-6.	<b>A</b> 1			-4.48	d		+1.4	4 0				+0.24					- 37.1	<b>b</b> +21.0			+3.				0.656		1-40.
	1600			-790.								-8.27	d+	14.50	+7.1	4 C		-5.	70 c	-8.8	bЪ					-974.0	a -90	5.7 C	1 +4.	00 0			0.000	<u>ч</u>	1-10.
										(g																									
														· .						Coba	lt-E	ase Al						-	1 -1.	01	-2940.0	a -2	6.6		1 -2.
	1675	100	200	1		+51.	3 4				T	-6.43			-4.4			Ι			1	-0.410			-194.0			7.22a			-2909.0			3	-2.
	1750	100	200			+47.						-6.37			-4.4	6						-0.210				a -56.9								a	1 -1.
	1900	100	200			+77.					-	-3.55	e		-2.5	j3 a						+0.20			-141.0		<u>q</u>	2.920	+19.	<u>9 a</u>	-2240.0		<u></u>	4	-2.
		100	200	+795.	ਰ ਰੱ						T				-2.0	57 c						+0.37					-1		<b>_</b>					+	-0.
	1900												T		-1.1					1		+0.82							<b>_</b>					-+	+1
		1000		+76.			74 4								+0.	533d		T				+0.16					_		+						-2
		1000		+1820.		+15.					-				+0.5	83d						+1.43											<b></b>		-0
		500		+484.			71d				-		_		+0.3	156-d						+0.16					_								1 +0
		500		+434.			60 d		-		-				+1.	5 d		1				+7.69							+						-1
		500		+878.		-52.									-2.1	23 d						-20.3					_		+			-+		-+	-5
		500		+579			84 d				-+		-+-		-10.										+16.6		_								+0
		500		+2531.							-		-		+1.			T		1		+1.61			-73.3				1					1-272.0	
	2125									-787.	0 c		-				+138.0	c +567.	0 0		. 1			5.35c		a +452.0									
	2227	<u> </u>	-	1 11150						161								+66		-		+1.72	a -	3.00a	-20.0	a +87.3	a		1					-51.3	a -2.

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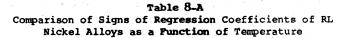
#### Table 7-A

#### Regression Coefficients for Maximum Penetration

													·····				1-1	
			Elen		· · ·							1.00			1			
Mn	Fe	Co	Ni	Cu	Y	Zr	СЪ	Mo	La	Ce	Hf	Ta	W	Re	R	SE	N	I
	1					L		<u></u>				••••••••••••••••••••••••••••••••••••••						
		Nickel-	Base Allo	ys '														
a		c -0.875 c			+76.5 c		-1.54c			+36.3 a	a	-0.943c			0.560		111	32.5
) a	-0.445	c -0.629 c			+170.0 c		+2.98c			+49.4 2	2	-0.002c		-	0.562		111	74.3
a	-0.055	c -0.670 c			+33.5 c	+127.0 c			a	+23.6	2	+3.08 a		1	0.752		108	75.0
, c		-0.008 c					-320.0 c		d				+10.6c a	-	0.95		20	145.0
' C		+3.95c c					-2952.0 c		c				-8.49			35.7		1742.0
) a	1	-0.262 c					-2522.0 c	+38.3	d				+9.20			15.7		1280.0
a	1	+0.445 c	1		1	1	-1511.0 c		d				-0.795			26.3	16	
' a!	1	-0.308 c					-1925.0 c		Ы				-1.40 0	-		18.3	23	1012.0
) 3	1	+0.0006c	1				+3054.0 c	+17.6	c				-0.940		0.92	17.3	12	-1422.0
' c	-39.7	c +2.97 a			+67.3 0	+705.0 c	1	-3.55	c -5.15	C	1	+9.72 c	-9.91	à	0.97	0.98	1 31	-265.0
c	-3889.7	c +3.05 c			+232.4 c	+2731.3 c		-14.1	c -51.7	C		+62.4 c	-174.7		1.00	0.94	1 13	-640.7
i c	-40.0	c +1.22 c			-29.7 c	+151.0 c	+4.56c	-1.00	-2.36	C		-2.97 c	-1.11	+0.2000	0.68	7.40	46	13.8
, c	+53.1	c +0.780 c			-54.3 c	+95.6 c	+3.95c	+2.47	-0.96	2c		-0.831c		+2.00	0.71	7.49	42	-63.4
1 9	+7.27	c +0.249 c			-37.1 b	+21.0 c	-1.32c	+3.89	c -0.31			-0.374c	+1.61	+2.83	0.97	2.47	25	5.6
c	-5.70	c -8.836 b	3			-974.0 a	-98.7 c	1 +4.06	c	-0.656	d	-40.4 c	+2.45		0.91	29.9	42	-197.0
															,			
		Cobalt-	Base Allo											-	1			
a	Ι		-0.410c		-194.0 a					a -26.6 a	a	-2.15 a				13.4	48]	127.0
) a			-0.2100		-193.0 a		-6.64a		-2909.0	a -25.8 a	a	-2.05 a	-2.05	3	0.86		48	126.0
5 a			+0.205		-141.0 c		-2.9 <b>3</b> c	+19.9	-2248.0	c -13.1 c	c	-1.27 c		c		21.7	49	76.5
' c			+0.3760		-10.4 c		1	1	1		1	-2.75 c		9	0.94	3.67	111	24.6
i c			+0.324 0		-27.0cc		1	L	1			-0.318c		d	0.57	5.95	15	-64.9
53c 13c 56c	I		+0.1630		-43.6 a			1	1		1	+1.91 b		4	0.99	1.29	11	-25.9
130			+1.43 a		-41.1 a		1	L	1		1	-2.06 b			0.85	3.31	28	-26.8
;6c	1		+0.163 c		+3.930		1	[				-0.647c		d	0.97	0.727		0.100
; c			+7.69 c		-7.300	1						+0.131c			0.98	0.67	· · · · · · · · · · · · · · · · · · ·	-132.0
5 d			-20.3 c	-3.23c	-47.0 c	1					1	-1.38 c	-0.7910	4	0.96	3.04	16	+318.0
c			-44.2 c	-7.46c	+16.6 c						1	-5.29 c	-14.3	9	0.998		8	+931.0
5 b	1	1	+1.61 a	-3.39b			1				1	+0.010c		a	0.97	2.61	20	-98.5
c +138.0c		с	+4.43 c	-5.35c		+452.0	1				-272.0c			-	0.95	3.66	22	+58.7
19 cl +26.9a	+66.5	c	+1.72 a	-3.00a	-20.0.a	+37.3 a	3				-51.3a	-2.05 b	+1.37		0.82	3.16	42	-17.8

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n 60% but less than 90% of tests per regression analysis, I - Intercept value for regression equation



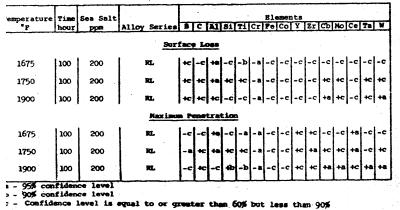
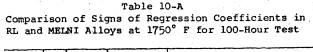


Table 9-A Comparison of Signs of Regression Coefficients of MELNI Alloys as a Function of Temperature

emperature °F	time hour	Sea Salt	Alloy Series	Elements BCALTICr Pe Co Y Zr Ch Mo La Ta W Re
			Sur	face Loss
1600	1000	5	MELINI	+c       +c       +c       +c       +c       +c       +c       +c       +b       +c         +c       -c       +c       +c <t< td=""></t<>
1800	1000	5	MELNI	+c -c +c +c -c +c +c +c +c +c +c +c +c +c
			Maximu	m Penetration
1600	1000	5	MELNI	-c +c -c -c +c +c +c -c +c +c -c +c +c +c +c
1800	1000	5	MELIKI	<b>4c -c -b -c +c +c +c -b +c -c +c -c -c +c +c</b> +c



perature °F	Time hour	Sea Salt ppm	Alloy Series	B	с	A1	Ti	E Cr	-	nent Co	_	Zr	Мо	Ta	Ŵ
			Surface	Losi	5										
1750	100	200	RL Nickel	+c	+¢	+a	+c	-a	-c	-c	-c	-0	+c	+0	+c
1750	100	200	MELNI	+c	-c	+c	+c	+c	-c	+a	+b	+c	-b	+c	-c
			Maximum Pen	etra	atic	<u>on</u>					•				
1750	100	200	RL Nickel	-a	+c	+a	+c	- a	-c	-c	+¢	+a	+c	-c	+c
1750	100	200	MELNI	+c	-c	÷c	+c	+c	-c	+a	+c	+c	-c	+c	-c

· 90% confidence level

. Confidence level is equal to or greater than 60% but less than 90%

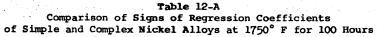
Table 11-A Comparison of Signs of Regression Coefficients of SLNI 1-13 at 1750° F as a Function of Exposure Time

rature	Time	Sea Salt								Ele	mer	ts					
P	hour	ppm	Alloy	Series	B	C	A1	Ti	Cr	Fe	Co	Y	Zr	Мо	La	Ta	W
			· · ·	Surfac	e Lo	288											
50	100	200	ME	LNI	+c	-c	+c	+c	+c	- C	+a	+b	+c	-Ъ	-c	+c	-c
'50	1000	5	MES	LNI	+c	-c	+c	+c	+c	-c	+c	+c	+c	-c	-c	+c	-c
			Ma	kimum P	ene	tra	tio	<u>1</u>									
'50	100	200	ME	LNI	+c	-c	+c	+c	+c	-c	+a	+c	+c	-c	-c	+c	-c
50	1000	5	ME	LNI	+c	-c	+c	+c	+c	-c	+c	+c	+c	-c	-,c	+c	-c

15% confidence level

Confidence level is equal to or greater than 60% but less than 90%

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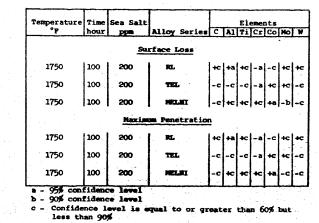
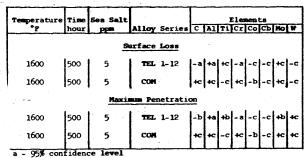


Table 13-AComparison of Signs of Regression Coefficientsof TEL Alloys and Commercial Alloys at 1600° F for 500 Hours



b - 90% confidence level

c - Confidence level is equal to or greater than 60% but less than 90%

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#### Table 14-A

Comparison of Signs of Regression Coefficients of RL Cobalt Alloys as a Function of Temperature

	Time	Sea Salt						1	Eler						
* <u>*</u>	hour	ppm	Alloy Series	C	Ťi	Cr	Øi	Y	Zr	Cb	Mo	La	Ce	Ta	Ň
			Surface	Los											
1675	100	200	FL.	+6	a		-c	-*	- 2	-a	-c		- a		-
1750	100	200	RL	+c	- 8	- a	-c	- a	- a	- 8	+b		- a	-a	-
1900	100	200	<b>R1</b> .	+c	-c	- a	+c	-c	- c	-c	+#	-c	-c	-c	+
			Naxiana Pen	etra	ntid	on '									
1675	100	200	RL.	+=	-•	-a	-c	-a	- a		-c	-a	-a	-a	-
1750	100	200	. HL	+10:			-c	- 8	- 8	- 2	+ib	-a	- a		-
1900-	100	200	RL	+c	-c		+c	-c	-c	-c	+8	-c	- C	-c	+

90% confidence level

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Confidence level is equal to or greater than 60% but less than 90%

#### Table 15-A Comparison of Signs of Regression Coefficients in Simple and Complex Cobalt Alloys at Two Temperatures in 100-Hour Tests

		Ca	rbon		
	Surface Loss			num ation	Concentra- tion Range
Alloys	1750° F	1900° F	1750° F	1900° F	w/o
Simple	+c	+c	+b	+c	0.01-0.45
Complex	+c	+c	-c	+c	0.23-0.42

		Chro	mium		
	Suri	tace IS	Maxin Penetra	Concentra- tion Range	
Alloys	1750° F	1900° F	1750° F	1900° F	₩/0
Simple	-8	-a	-a	-a	15. <b>0-3</b> 5.0
Complex	+c	+c	-c	-c	27.8-34.7

		Nic	ckel		
	Suri		Maxis Penetra		Concentra- tion Range
Alloys	1750 1	1900° P	1750° F	1900° F	¥/0
Simple	-¢	+c	-c	+c	0-10.2
Complex	+c	-c	+c	+c_	0.5-10.7

		Ytt:	rium		
1	Surl		Maxis Penetra		Concentra- tion Range
Alloys	1750 P	1900° F	1750° F	1900° F	w/o
Simple	-a	-c	-a	-c	0-0.150
Complex	-c	-a	-c	-c	0-0.190

		Tani	talum		
	Surf		Maxim Penetra	Concentra- tion Range	
Alloys	1750° F	1900° F	1750° F	1900° F	¥/0
Simple	-a	-c	-a	-c	0-15.0
Complex	-c	+c	-c	c	0-4.0

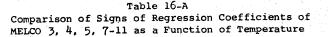
			Tung	isten		
		Surf		Maxir Penetra		Concentra- tion Range
1	Alloys	1750° F	1900° P	1750° F	1900° F	w/o
	Simple	-a	+c	-a	+c	0-15.0
	Complex	+c	+c	-c	+ċ	6.30-8.25
	Complex	40			1	0. 000.2

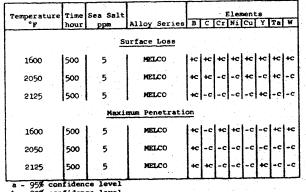
Note: Simple alloys are RL 10-15, 52-64 Complex alloys are MELCO 1-11

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b - 90% confidence level

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c - Confidence level is equal to or greater than 60% but less than 90%

Table 17a-A Comparison of the Regression Coefficients of Boron MELCO 1-11 at Two Temperatures and Two Test Conditions\*

. Г	Surfac	e Loss		enetration
T	1750° F	1900° P	1750° F	1900° P
100 hours	+465.0	+556.0	+796.0	+1094.0
200 ppm	С	с	c	c
1000 hours	-168.0	+445.0	+76.0	+1820.0
5 ppm	c	c	c	a

Cencentration range of B is 0% to 0.015%.

Table 17b-A Comparison of the Regression Coefficients of Carbon in MELCO 1-11 at Two Temperatures and Two Test Conditions\*

1	Surface	e Loss	Maximum P	enetration
[	1750° F	1900° F	1750° F	1900 F
100 hours	+1.30	+8.46	-10.4	+29.2
200 ppm	° C	C i	c	C _
1000 hours	-8.01	-15.1	-7.74	+15.6
5 ppm	, c	c	c	c

\*Concentration range of C is 0.23% to 0.42 %.

#### Table 17c-A

Comparison of the Regression Coefficients of Chromium in MELCO 1-11 at Two Temperatures and Two Test Conditions\*

· · [	Surfac		Naximus P	enetration
	1750° F	1900 F	1750° P	1900 P
100 hours	+0.213	+0.417	-2.67	-1.14
200 ppm	c	c	c	c
1000 hours	+0.021	-0.942	+0.533	+0.583
	c	ъ	c	c

\*Concentration range of Cr is 27.8% to 34.7%.

#### Table 17d-A

Comparison of the Regression Coefficients of Nickel in MELCO 1-11 at Two Temperatures and Two Test Conditions\*

	Surface	Loss	Maximum P	enetration
	1750° F	1900° F	1750° F	1900° P
100 hours	+0.544	-0.207	+0.376	+0.824
200 ppm	с	с _	c	c
1000 hours	+0.080	+1.30	+0.163	+1.43
5 ppm	с	a	с	a

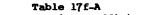
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Table 17e-A
mparison of the Regression Coefficients of Copper
LCO 1-11 at Two Temperatures and Two Test Conditions*

	Surface	e Loss	1 1	Maximum Penetrat				
	1750° P	1900 F	1	1750 7	1900 7			
100 hours	-2.12	+0.058		-2.87	-1.35			
200 ppm	c	c		с	c			
000 hours	-0.177	-3.55		+1.43	-2.87			
5 ppm	с.	-		c	ъ			

oncentration range of Cu is 0% to 2.95%.



mparison of the Regression Coefficients of Yttrium LCO 1-11 at Two Temperatures and Two Test Conditions\*

	Surfac	e Loes		Raximum Penetration				
	1750 P	1900° P		1750 2	1900 2			
100 hours	-0.823	-33.0		-10.4	-27.0			
200 ppm	c	•		c	c			
i000 hours	+5.01	+25.6		-43.6	-41.1			
5 ppm	c	Ъ		•	2			

Concentration range of Y is 0% to 0.190%.

			Table 17g-A
mparison	of	the	Regression Coefficients of Tantalum
LCO 1-11	at	Two	Temperatures and Two Test Conditions*

Surface		Maximum 1	Penetration
1750° F	1900° F	1750 1	1900 7
-1.66	+0.437	-2.75	-0.318
c	c	c	c
+0.077	-2.81	+1.91	-2.06
c	а	ъ	ъ
	1750° F -1.66 c +0.077	-1.66 +0.437 c c +0.077 -2.81	1750* F         1900* F         1750* F           -1.66         +0.437         -2.75           c         c         c           +0.077         -2.51         +1.91

Concentration range of Ta is 0% to 4.00%.

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Table 17h-A Comparison of the Regression Coefficients of Tungsten in MELCO 1-11 at Two Temperatures and Two Test Conditions\*

Г	Surface	a Loss	Maximum	Maximum Penetration		
	1750° P	1900 F	1750 F	1900 *		
100 hours	+1.41	+0.965	-0,654	+2.77		
200 ppm	C	c	c	c		
1000 hours	+0.001	-2.83	+2.79	+0.460		
5 ppm	с	•	c	c		

\*Concentration range of W is 6.30% to 8.25%.

#### Table 18-A Hot-Corrosion Results for Factorial Alloys<sup>1</sup> (Nickel-Base, 100 Hours of Test Operation)

		Chem			(2) Maximum Penetration, mils							,	
111.		Allo		•		1675 7			1750° r			1900 1	-
Alloy		<b>\$</b>	-					-	Test2				
Ho.	Wi	Cr	AL	Ti	2	3	4	7	8	2	16	B	14
44	Bal	15.0	2.5	4.5	54.6	<b>97</b> .9	10.7	36.6	39.9	50.1	34.9	<b>55</b> °5	58.6
45	Bal	15.0	2.5	9.0	8.6	8.9	7.5	15.3	51.1	11.4	11.5	15.5	21.6
46	Bal	15.0	5.0	4.5	15.8	>130.0	5.9	10.8	>130.0	43.7	46.7	21.5	63.3
47	Bal	15.0	5.0	9.0	7.7	9.9	6.5	9.4	>130.0	8.9	44.5	15.5	56.4
48	Bal	20.0	2.5	4.5	12.5	18.5	9.0	11.5	17.9	12.6	17.0	12.4	33.4
49	Bal	20.0	2.5	9.0	7.1	7.0	6.6	9.6	8.0	8.7	13.6	16.9	14.1
50	Bal	20.0	5.0	4.5	8.6	7.7	7.9	10.0	10.4	28.7	27.1	28.3	45.5
51	Bal	20.0	5.0	9.0	6.6	5.4	6.1	3.2	>130.0	8.9	31.1	ì1.5	14.4
1	1		ł	1	1		1	1					1

<sup>1</sup>Temperature - as given; Time - 100 hours; Fuel - diesel (1% sulfur); Air/Fuel - 30/1; Sem Salt - 200 ppm of air; Specimen Size - approximately 0.130 inch in dimmeter by 1.25 inches in length.
<sup>2</sup>A maximum penetration >130 mils indicates that the specimen has been corroded all the way through.

<sup>3</sup>Each test number represents a separate test run in which one specimen of each alloy is placed in a hot corrosion test chamber, which is then operated under specified conditions. Bal - Balmace.

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#### Table 19a-A Analysis of Variance for Hot-Corrosion Results for Factorial Alloys :1-Base, 100 Hours of Test Operation, All Data Included)

	Degrees		
Sums of	of	Mean	
Squares	Freedom	Squares	'F' Ratio
2378.39	2	1189.28	1.45
6367.56	1	6367.56	7.78*
1833.15	1	1833.15	2.24
2704.80	1	2704.80	2.31
53998.18	<u>66</u>	818.15	
67282.08	71		
	2378.39 6367.56 1833.15 2704.80 53998.18	Sums of Squares         of Freedom           2378.39         2           6367.56         1           1833.15         1           2704.80         1           53998.18         66	Sums of Squares         of Freedom         Mean Squares           2378.39         2         1189.28           6367.56         1         6367.56           1833.15         1         1833.15           2704.80         1         2704.80           53998.18         66         818.15

ficant at the 1% level.

Table 19b-A
Analysis of Variance for
Hot-Corrosion Results for Factorial Alloys
(Nickel-Base, 100 Hours of Test Operation
Test Runs 3, 8, and 13 Eliminated)

Sums of Squares	Degrees of Freedom	Mean Squares	'F' Ratio
2033.32	2	1016.66	12.31*
826.75	1	823.75	10.00*
19.93	1	19.93	<1.00
1007.06	1	1007.06	12.24*
<u>1438.99</u>	<u>18</u>	82.72	
5376.05	23		
	Squares 2033.32 826.75 19.93 1007.06 <u>1438.99</u>	Sums of Squares         of Freedom           2033.32         2           826.75         1           19.93         1           1007.06         1           1438.99         18	Sums of Squares         of Freedom         Mean Squares           2033.32         2         1016.66           826.75         1         323.75           19.93         1         19.93           1007.06         1         1007.06           1438.99         18         82.72

ficant at the 1% level.

The following test results have been averaged in Table 19b for each alloy: Tests 2 and 4, 7 and 9, and 12 and 14.

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#### Table 20-A 小孩子 生化化学

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#### Hot-Corrosion Results for Factorial Alloys<sup>1</sup> (Cobalt-Base, 100 Hours of Test Operations)

		Chemical Maximum Permetration Composition of 1675 F, 1750 F							1900 F	
DISCO Alloy	Alloys		Tes	, 1 <sup>(2)</sup>	Test	s <sub>(5)</sub>	·	• 3 <sup>(2)</sup>		
No.(2)	Co	Cr	Wi	Ta	A	B	λ	В		
1	Bal	25	5	6	3.9	2.8	6.5	8.6	14.7	16.6(3)
2	Bal	30	5	6	5.3	3.8	6.8	9.2	14.9	18.0
3	Bal	35	5	6	5.4	8.3	8.2	9.9	13.9	16.1
-4	Bal	25	15	6	4.8	5.0	7.2	7.6	14.3	17.2
5	Bal	3,0	15	6	6.8	5.8	11.3	8.6	15.6	17.1
6	Bal	35	15	6	6.6	8.9	8.9	12.4	14.8	16.2
. 7	Bal	25	25	6	5.9	5.0	8.4	9.6	16.5	38.7
8	Bal	30	25	6	6.6	5.1	10.7	11.8	19.6	19.7
9	Bal	35	25	6	6.8	10.3	12.1	13.6	16.1	20.7

LTemperature - as given; Time - 100 hours; Fuel - diesel (1% sulfur); Air/Fuel - 30/1; Sea Salt - 200 ppm of air; Specimen Size - approximately 0.130 inch in diameter by 1.25 inches in length.

<sup>2</sup>DISCO 1A to 9A - Specimens are used, as cast. DISCO 1B to 9B - Specimens are heat treated, 1200° C (2192° F), 24 hours, WQ. Heat-treated (B) and nonheat-treated (A) specimens of each alloy were tested at one time, under a specified temperature condition. <sup>3</sup>Estimated value.

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#### Table 21a-A Analysis of Variance for Hot-Corrosion Results for Factorial Alloys (Cobalt-Base, Overall Factors)

Source of Variance	Sums of Squares	Degrees of Freedom	Mean Squares	'F' Ratio Line	Line
DISCO, A VS B(H)	18.96	1	18,96	1/5 = 3.09	1
Temperature (T)	1097.21	2	548.60	2/5 = 89.35*	2
Chromium (Cr)	43.34	2	21.67	3/6 - 3.74	3
Mickel (Mi)	54.72	2	27.36	4/8 = 28.21*	4
HxT	12.28	2	6.14	5/8 - 6.33-	5
H x Cr	14.26	2	7.13	6/8 = 7.35*	6
T x Cr	26.11	•	6.53	7/8 = 6.73=	7
Residual		28	0.97		8
Total	1303.81	53	-		

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#### Table, 21b-A. Ser Analysis of Variance, for Ser Corrosions Results: for Factorial: Alloya (1996) (Cobalts-Base): Over 3-Temperature, Levels)

Source at of Variance	Sume of	Degrees of Freedom	Nean	IP' Ratio	Line
1	Temperature - 1675" F				
DISCO, A vs. B(H) Chromium (Cr) Sickel. (Nk) H x Cr Residum2 Total	0 40.34 8.68 19.12 4.79 72.93	1222	0 20.20 4,34 9.56 0.43	1/4 = 0 2/4 = 2.11 3/5 = 19.04* 4/5 = 19.91*	12345
	<u>.</u>	mperature -	1750° F		
DISCO, A va B(H) Chromium (Cr) Nickel (Ni) Residual Total	6.95 25.06 24.41 <u>15.35</u> 72.27	1 2 2 <u>1</u>	6.95 12.53 12.20 1.32	$1/4 = 5.27^{**}$ $2/4 = 9.49^{*}$ $3/4 = 9.24^{*}$	1 2 3 4
	T	emperature -	1990° F		
DISCC, A vs B(H) Chromium (Cr) Nickel (Ni) Residual Total	24.22 4.01 27.13 5.99 61.35	1 2 12 17	24.22 2.00 13.56 0.50	1/4 = 48.44* 2/4 = 4.00** 3/4 = 27.12*	1 2 3 4

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ESCRIPTIVE NOTES (Type of report and inclusive dates) Search and Development		
THORIS (First name, middle initial, last name)		
Field, D. J. Fisk, and H. von E. I	Doering	
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BETRACT	

The use of gas turbines in marine power plants depends in part on the velopment of superalloys which not only possess high temperature chanical properties but also resist the corrosive effects of sea salt, part of a program to develop such alloys, a total of 137 experimental d commercial superalloys, both nickel and cobalt based, were exposed : burner rigs where controlled amounts of sea salt were added to the mbustion products of sulfur-containing diesel fuel. Test temperatures inged from 1600° to 2125° F. Times ranged from 86 to 100 hours with 10 parts per million, and from 489 to 1100 hours with 5 parts per .llion salt. Corrosion was measured by recording both surface loss d maximum penetration. This experimental work was performed by the meral Electric Company under contract to the Naval Ship Research and velopment Center. For each group of alloys tested under similar contions, a linear regression equation was found that shows the average intribution of each alloying element to the amount of corrosion. The fects of the alloying elements were found to vary with changes in mperature, salt concentration, and whether or not the particular ement was part of a simple binary or tertiary alloy, or a complex

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Analyses of variance methods were applied to two sets of facly designed compositions, one of nickel-base alloys and one of -base alloys, to determine the possible significance on ion of various proportions of single elements and interactions elements. It was found that in the cobalt alloys significant ctions existed between heat treatment and temperature as well tween heat treatment and chromium content.

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